A STUDY INTO THE FATTY ACID COMPOSITION AND THE PHYSICO-CHEMICAL INDICES OF BONE FAT AND METHODS FOR IMPROVING ITS QUALITY

A high fat content in bones allows to utilize them as a raw material for the production of fats having different physico-chemical characteristics.

There is scanty information characterizing differential fatty acid composition and physico-chemical indices of bone fat as related to the nomenclature of bone from which fat is produced.

Bone fat, produced according hygienic rules, is a standardquality product having milder consistency as compared to beef and mutton tallow. It contains essential fatty acids, has a pleasant specific odour and taste and a yellow colour determined by the availability of vitamin and provitamin A (carotene), is easily digested (97%) /1, 2/.

An extensive study was carried out, aimed at determining the fatty acid composition of the fat produced from various skeletal bones of cattle and pigs, viz. of vertebrae, epiphyses, pelvic bones and ribs of cattle and of long bones of pigs.

Bones were ground into pieces no more than 40 mm, defatted at 90°C, re-ground down to 25 mm and centrifuged.

The fat after primary and secondary defatting was purified and bleached in a separator and sampled for chemical analyses.

Fatty acid composition was determined by gas-liquid chromatography with preliminary methylation of the samples by A.G.Vereshtchagin's method /3/. Fatty acid methyl esters were separated with a "Pye" chromatograph (England): carrier gas - argon; detector - argon, ionization; hard carrier - Celit-545 (100-120 mesh); liquid phase - polyethylene glycol-adipate (15% of carrier weight). They were identified by the values of relative retention volumes.

Results are given in Table 1.

In all the samples of bone fat, produced from different bones, octadecaenoic acid prevailed (from 38.1% in the fat from beef ribs to 52.6% in the fat from pig long bones). Essential fatty acids content is comparatively low, in fat from pig bones it was

2.3-5.8%.

).

9

We established that the level of octadecanoic acid was within 11.7-19.4%, it being least in the fat from pig long bones (9.7%); this is determined with a lower melting temperature and a higher iodine number.

Of saturated acids, hexadecanoic acid content is highest (22.3-31.2%), in pig bone fat it is 22.3-23.5%.

In all the cases the fat after secondary defatting of bones has a higher iodine number.

The comparison of the fatty acid compositions of various bone fats allows to determine from what kind of bones, mixed bones or fats mixture a given fat was manufactured.

The following correlations are suggested and calculated (Table 2):

 $R_{1} = \frac{\text{unsaturated acids } C_{14}}{C_{14}(\text{tetradecanoic})} \times 100$ $R_{2} = \frac{\text{total acids } C_{14} + \text{total acids } C_{15}}{C_{16}(\text{hexanoic})} \times 100$ $R_{3} = \frac{C_{18} \text{ ll}}{C_{16}} \times 100$

Simultaneously, the basic physico-chemical characteristics of fat differentiated by bone anatomical features (Table 3).

Total unsaponifiables in bone fats are significantly higher than in other animal fats produced from soft fatty materials. Lecithine content in bone fat is 4-6 times as high as in fat produced from soft raw materials. Bone fat has good emulsifying capacity.

Dilatometric studies (Fig. 1) showed that fat from epiphyses, mixed beef bones and pig long bones is of the best plasticity as it contains 15-30% hard triglycerides at 20°C.

Bone fat quality is greatly determined with processing conditions and the equipment used for extraction. The analysis of the available methods for bone defatting indicates that they are time-consuming, accompanied with a considerable loss of proteinaceous substances, the finished product gives low yields and is of low quality, this being conditioned by the use, in most cases,



Fig. 1. Hard triglycerides content in various bone fats as influenced with remperature

- 886

1



Fig. 2. Kinetic ourves of bone fat oxidation: 1 - fat produced by means of autoclaving; 2 - fat produced by the developed technology of batch-type apparatuses. Procedures of continuous bone defatting by means of impulse methods require great consumption of water, are connected with protein losses and with complicated separation of fat from water due to fat emulsification.

In the All-Union Meat Research Institute (VNIIMP) a rational complex technological process has been developed, which provides fast continuous two-stage defatting of bone with simultaneous dehydration of the latter at mild temperatures /5/. The process consists in the dry heating of thin-layer ground bones at 80-85°C for 10-12 min., in the immediate removal of fat from the heating zone, in bone re-defatting in the centrifugal field for 3-4 min. and in 30 min. drying.

The above method provides satisfactory sanitary production conditions and high yields of fat (85%), prevents the decomposition and losses of hone protein fraction; due to it, the yheld of the latter after drying is 49-50% of the initial raw material weight.

Fat resistance to oxidative destruction as influenced with bone defatting conditions was studied.

Samples of the fat produced from mixed bones by the developed method and by autoclaving at 130°C for 3.5 hrs were comparatively studied by means of kinetic oxidation at 98°C; at the same time, their oxidative changes were followed during extended storage (13 months) at -8 to -10°C.

Oxidation curves in Fig. 2 shows that the rate of iodine number growth of the fat produced at moderate temperatures is twice as less as compared to fat produced under very high temperature conditions. Storage for 6 months did not affect the characteristics of fat samples, manufactured by the method developed, whereas it did deteriorate the sensoty qualities of fat, produced in autoclaves.

High temperatures cause the destruction of natural antioxidants, this being accompanied with a reduced stability of bone fat to oxidation, thus, carotene content in autoclaved bone fat was 0.21 mg%, while in fat produced by the new method was 0.33mg/ /6/.

Moderate temperatures and a short processing time prevent lipids and lipoids decomposition, this providing high qualities of bone fat and permitting its utilization in various branches of industry.

LITERATURE

- I. Грау Р. Мясо и мясопродукты. М., Пищевая промышленность, 1964.
- 2. Либерман С.Г., Петровский В.П. Справочник по производству животных жиров. М., Пищевая промышленность, 1972.
- ³. Верещагин А.Г. "Биохимия", 5, 28, 1963.
- 4. Wolff J.P., Audian F. Rev. Franc. Corps., 11, 1966, 17.
- ⁵. Либерман С.Г., Фнйвишевский М.Л. Теоретическое обоснование новой технологии комплексной переработки кости. "Тр.ВНИИМПа", ХХП. М., 1970.
- 6. Цимбалова Н.М. Влияние способа производства на кератин костного жира. "Мясн.индустр.СССР", IO, 1969.

Table 1

Fatty acid composition of bone fats

	Kind of bone										
Chargcteris- tics	pig long beef ribs bone				beef bee pelvic epi bone sis		beei epij sis	f ohy-	mixed epipl and v brae(50%)	nixed epiphyses and verte- brae(50%/ 50%)	
]	Def	at	t 1	ng	an dhug daib dig vaa g			
	lst	2nd	lst	2nd	lst	2nd	lst	2nd	lst	2nd	lst
Content of acids, %:		3	4	2	6	1	8	9	10	11	IE
tridecenoic C ₁₃ /I/		Tr	ac	8 5					022	Trace	95
tetradecanoic	115	155	104	200	27	10	20	163	101	140	22
tetradecenoic C ₁₄ /I/	112	013	0.75	103	0.9	0,8	0,6	0.92	129	072	0.9
pentadecanoic C ₁₅ /9/	015	010	0,62	097	0,8	0.7	0,6	049	0,89	0.53	07
pentadecadie- noic C15/II/	0.03	Trace	es025	0.38	02	01	01	020	0.69	021	02
hexanoic C16/0	0/2231	2351	3120	39,47	244	244	237	2522	2275	2586	204
C ₁₆ /I/	445	532	397	361	37	44	28	361	5.57	3.68	43
heptadecanoic C ₁₇ /0/	0,69	046	133	167	12	13	10	0.86	137	137	0.8
heptadecenoic C ₁₇ /I/	0.79	049	072	0,84	10	0,8	07	0,61	128	0.95	0.6
octadecanoic C ₁₈ /0/	970	975	17.00	1828	152	165	140	1429	1170	1378	194
octadecaenoic C ₁₈ /I/	5030	5264	3952	3813	482	464	517	4978	4591	4816	412
octadec adieno : C18/II/	1014	605	369	282	23	28	27	238	478	308	30
octadecatrien C ₁₈ /III/	018		T	r	a. 0	•	s		106	Q26	Tra- ces
eucosenoic C ₂₀ /I/	011		T	r	a 0	e	8		0,65	r Tra	ces
Total	10000	10000	9999	100,00	10000	10000	999	9990	10000	10000	10000

<u>1</u> <u>2</u> <u>3</u> <u>4</u> <u>5</u> <u>6</u> <u>7</u> <u>8</u> <u>9</u> <u>10</u> <u>11</u> <u>12</u> Iodine number, <u>6</u> J <u>6240</u> <u>6240</u> <u>4410</u> <u>4460</u> <u>488</u> <u>496</u> <u>521</u> <u>5230</u> <u>4920</u> <u>5060</u> <u>436</u> Melting point, <u>°C</u> <u>3950</u> <u>4070</u> <u>-</u> <u>-</u> <u>410</u> <u>435</u> <u>410</u> <u>410</u> <u>425</u> <u>415</u> <u>460</u>

xnot identified

,

,

,

Table 2

Bone	fot complex	Correlations, %							
DOHe	Ter Somptes	R ₁	$R_1 R_2 R_3 S=R_1+R_2$		R ₁ /R ₂				
Beef	fat from bones:								
	pelvic	42.8	16.4	9.4	59.2	2.6			
	epiphysis	30	13.5	11.4	43.5	2.2			
	vertebrae	40.9	14.9	11.2	55.8	2.8			
	ribs	40.7	11.1	11.8	51.8	3.6			
	mixed (50% ver- tebrae and 50%	67 E	21 0	01.0	00 F	2.0			
Dat .	epipilyses)	07.02	21.0	21.0	88.7	3.2			
bone	s	97.4	10.9	45.4	108.3	8.9			

Table 3

Champataniation		Bones					
	UNIC	mixed	epiphyses	pig long			
Hardness	g/cm	300	226	71			
Hardening tempera⊷ ture	°c	31.2	28.1	23.3			
Melting point	°c	41.8	41.0	39.5			
Lecithine content	96	0.147	0.131	0.176			
Iodine number	% J	50.3	52.2	62.4			
Unsaponifiable sub- stances	%	0.73	0.70	0.86			