XXI. EUROPAISCHER KONGRESS DER FLEISCHFORSCHUNGSINSTITUTE

ALLUNIONS-FORSCHUNGSINSTITUT DER FLEISCHINDUSTRIE DER UASSR

STRUKTURELL-MECHANISCHE BIGENSCHAFTEN VON KOLLAGENLÖSUNGS-PRODUKTEN

W.M.Gorbatow, O.O.Bablojan, P.M.Golowanowa, G.E.Limonow, 0. P. Borowikowa

ZUSAMMENFASSUNG

Es wurden die Kollagenlösungsprodukte untersucht, die nach U_{kall} -Salz-(PRK) und fermentativer Behandlung (FRK) von kollagen Laltigen Rohstoffen der Fleischindustrie (Schnen) gewonnen wurden.

Es wurde die Abhängigkeit zwischen der Viskosität von PRK und FRK und der Eiweißkonzentration, der Veränderung von pH des Mediums sowie der mechanischen Einwirkung festgestellt.

- Die pH-Anderung bei FRK von 3,10 bis 8,60 ruft die Viskositätsabnahme um fast das zweifache hervor, während die gleiche pH-Anderung bei PRK zur unwesentlichen Senkung des Ausgangs-Wertes führt.

- Bei den 30 Minuten bei 40°C denaturierten und nachfol-Send abgekühlten PRK und FRK sinkt die Viskosität im saueren PR-Gebiet fast bis zur Viskosität des Lösungsmittels und bleibt bei der weiteren Einwirkung auf das System von Soherspannungen Praktisch ohne Veränderungen.

- Im alkalischen pH-Gebiet übertrifft die Viskosität des TRE-Systems 1,5 Stunden und des PRK-Systems 20 Stunden nach der Wirmedenaturierung und nachfolgender Abkühlung die Viskosität der Ausgangsprodukte der Lösung, was mit der Bildung einer vollunsgangsprodukte der Losung, was mat der bestehenen Sommeneren Struktur aus Denaturierungsprodukten wahrscheinlich Verbunden ist.

Die erhaltenen Ergebnisse können bei der Herstellung von künstlichen Kollagenmaterialien (Wursthüllen, Fleischwarenüberzügen, künstlichen Kollagenfaden u.a.m.) ausgenutzt werden.

XXIst EUROPEAN MEETING OF MEAT RESEARCH WORKERS

ALL-UNION RESEARCH INSTITUTE OF THE MEAT INDUSTRY

STRUCTURAL-AND-MECHANICAL PROPERTIES OF COLLAGEN SO-LUBILIZATION PRODUCTS

XX

V.E.Gorbatov, O.O.Babloyan, P.M.Golovanova, G.E.Limonov, O.P.Borovikova

SUMMARY

Collagen solubilization products, resulting from alkali-Salt (PRC) and enzymatic (FRC) treatments of collagen-contai-Wing Wastes of meat production (tendons), have been studied.

Relations of PRC and FRC viscosity to protein content, Asedium pH and mechanical treatment have been established:

a change in FRC pH from 3.10 up to 8.60 reduces the A change in FRC pH from 3.10 up to 0.00 in the second ty by nearly two times, whereas a similar change in the by a second to any slightly; DH of PRC decreases the initial viscosity only slightly;

^{wing} in the acid range, the viscosity of PRC and FRC follo-down alexaturation at 40° C for 30 min. and cooling, decreases Acount almost to the viscosity of the solvent and further, pracically, does not change under shear stresses;

in the alkaline range, after heat denaturation and cooling, in the alkaline range, after near users of PRC in 20 hr to the viscosity of FRC in 1.5 hr and that of PRC in 20 hr. the viscosity of FRC in 1.5 hr and that the viscosity of the initial solubilization products, this $t_{\rm big}^{\rm todg}$ the viscosity of the initial solution of a more perfect $s_{\rm t_{\rm big}}$, obviously, being due to the formation of a more perfect

etructure from denaturation products. The effect of mechanical treatment upon PRC and FRC visco-

 d_{ied}^{-v} at various levels of solubilization products and d_{ied}^{-v} . It was found that PRC and FRC viscosity at the concentration of weak found that prc and FRC viscosity is the concentration of the second s $t_{\rm log}$ of up to 1% decreased sharply under mechanical treatment,

while at 1.5% this decrease was less intensive, this being, probably, connected with a more developed structural network in concentrated products.

The results obtained can be used in the production of artificial collagenous materials (sausage casings, coatings for meat products, man-made collagenous threads, etc.).

XXI ЕВРОПЕЙСКИЙ КОНГРЕСС РАБОТНИКОВ НИИ МЯСНОЙ ПРОМЫШЛЕННОСТИ

Всесоюзный научно-исследовательский институт мясной промыпленности СССР

СТРУКТУРНО-МЕХАНИЧЕСКИЕ СВОЙСТВА ПРОДУКТОВ РАСТВОРЕНИЯ КОЛЛАГЕНА

В.М.Горбатов, 0.0.Баблоян, П.М.Голованова, Г.Е.Лимонов, 0.П.Боровикова

КИДАТОННА

Исследованы продукты растворения коллагена, полученные после щелочно-солевой (ПРК) и ферментативной (ФРК) обработок коллагенсодержащих отходов мясной промышленности (сухожилий).

Установлена зависимость, связывающая вязкость ШРК и ФРК с концентрацией белка, изменением pH среды, механическим воздействием:

- изменение pH ФРК с 3, IO до 8,60 вызывает уменьшение вязкости почти в два раза, тогда как такое же изменение pH у ШРК приводит к незначительному снижению исходного показателя;

- в ПРК и ФРК в кислой области рН, подвергнутых денатурации при 40°С в течение 30 мин. с последующим охлаждением, вязкость снижается почти до вязкости растворителя и в дальнейшем практически не изменяется при воздействии на систему сдвиговых напряжений:

- в щелочной области рН после тепловой денатурации и последующего охлаждения, вязкость системы для ФРК через I,5 часа, и для ШРК через 20 час. превышает вязкость исходных продуктов растворения, что, по-видимому, связано с образованием более совершенной структуры из продуктов денатурации.

Исследовано влияние механического воздействия на вязкость ПРК и ФРК при различных концентрациях продуктов растворения, так при концентрации до I% она резко уменьшается, а при увеличении концентрации до I,5% - в меньшей степени, что, вероятн^о, связано с более развитой структурной сеткой в концентрированных продуктах.

Полученные результаты могут быть использованы в производст ве искусственных коллагеновых материалов (колбасных оболочек, покрытий на мясопродукты, искусственных коллагеновых нитей и др.).

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STRUCTURAL-AND-MECHANICAL PROPERTIES OF COLLAGEN SO-LUBILIZATION PRODUCTS

V.M.Gorbatov, 0.0.Babloyan, P.M.Golovanova, G.E.Limonov, 0.P.Borovikova

In the recent years a great number of papers appeared (1-5) on the use of collagen-containing materials in the production of artificial sausage casing, coatings for meat products, etc.

Collagen is most available of all the animal proteins. A growing interest in collagen is due to the fact that it can be relatively easily isolated in a pure form as soluble macromolecules and is of great chemical and thermal stability, as well as of a unique, highly ordered structure. Fibrous collagens can be treated chemically, mechanically, etc. Resulting solubilization products possess the properties close to those of native collagen solutions (pro-collagen) and retain the natural three-helical structure of macromolecules.

Peculiarities of diluted collagen solutions have been studied in detail (2), while the properties of concentrated so lutions are studied very little despite the fact that recently they have found commercial application.

This paper reports the results of a study into the rheolor gical properties of collagen solubilization products, which der termine the behaviour of the latter during the formation of film coatings, oriented fibers, etc. These data are of consider rable interest for elucidating the conditions under which collar gen solubilization products can be used as film-formers. X

2.

It is known (6, 7) that in disperse systems and solutions of high molecular compounds, there is a spatial structural Astwork which constitutes a loose carcass made of primary partioles, of their chains and aggregations resulting from inter-Rolecular interactions of structural elements with one another and with solvent molecules. Structuro-mechanical propertiand with solvent molecules. Schwart moth of the spatial structural network. The nature of deformation processes during flow of disperse or high-molecular systems is determined with relaration phenomena of destruction and reconstitution of the structures existing in the system, the former being dependent upon the peculiarities in the structure of polymer macromolecules.

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Collagen solubilization products are structured liquids Collagen solubilization products are structured with a relation of effective viscosity to the operating stress. In such liquids, every rate of deformation in Stationary flow has a corresponding degree of structure equilibrium breakage.

Experimental

^{Objects} of the study were collagen solubilization products Ubjects of the study were collagen Schutzer (1, 2), and products of enzymically solved collagen (FRC) obtained according to the VNIIMP method of collagen (beef Achilles tendons) digestion with pepsin.

The viscous properties of these products were studied at Protein concentrations from 0.8 up to 2.0%. For the experiments, Viscosimeter "Rheotest" Type RV (GDR) was used, which allows to measure viscosity within a broad range of shear rates. To ^{devisure} viscosity within a broad range of Smith ^{devive} flowability curves, shear rates were changed stepwise ^{from 0.3} up to 1.312 sec⁻¹. Preliminary tests indicated that, in case of shear deformation of collagen solubilization products, the $t_{\rm be}^{\rm vase}$ of shear deformation of collagen solutions $t_{\rm be}^{\rm vase}$ equilibrium between the structure being destroyed and that

being reconstituted was recorded by the end of the first minute of deformation. Therefore, in the experiments the deformation for any given shear velocity was continued for 1 minute.

The experimental data obtained allowed to find a relation of the effective viscosity of collagen solubilization products of different concentrations to a shear rate (Fig. 1). This relation can be described with the following equation (8):

$$ef = B^* \cdot \dot{\varepsilon}^{-n}$$

where B^* - effective viscosity at $\dot{\xi} = 1 \text{ sec}^{-1}$; $\dot{\xi}$ - deformation rate, sec.⁻¹;

1

m - rate of structure breakage, or flow index.

From Fig. 1 it is clear, that effective viscosity relations to shear rates within the forementioned range of rates in logarithmic co-ordinates are, practically, parallel straight lines. Thus, flow index is a constant for any type of solutions, for FRC it being 0.775 and for PRC 0.860. Therefore, we can compare the effective viscosity of the test solutions at the same deformation rate.

Fig. 2 shows changes in FRC and PRC effective viscosity as related to concentration. It is clear that for FRC this relation is less vivid than for PRC. E.g., at 1.5% concentration FRC viscosity is 3 times as low as that of PRC, this, possibly, being due to differences in the aggregation sizes of collagen solubilization products obtained by the two methods.

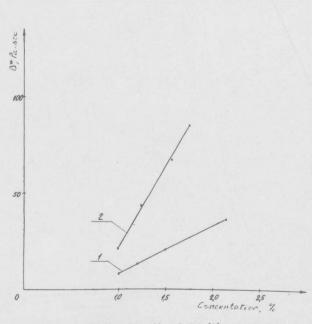
The viscosity of the solutions remains high, this preventing the preparation of equally dense films and fibrous materials. To decrease PRC viscosity, the authors studied the effect of pH, temperature and mechanical treatment. pH of FRC was changes from 3.10 up to 8.60 and that of PRC from 3.10 up to 8.30 with NH, OH solution. Figs. 3 and 4 demonstrate FRC viscosity falling nearly by twice and PRC viscosity changing only slightly. This, obviously, points to a higher deformability of

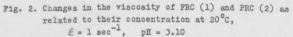
With changing pH of the medium.

 $t_{\rm he}$ collagen molecular rods of FRC and of their aggregations $m_{\rm 1+1}$

Z 3 9 5 6 7 8 5 m related to deformation rate at pH=3.10 and t=20°C ---- PRC concentration, \$; 1 - 1.00, 2 - 1.55; 3 -1.75

FRC concentration, \$; 4 - 1.00, 5 - 1.50, 6 -2.15





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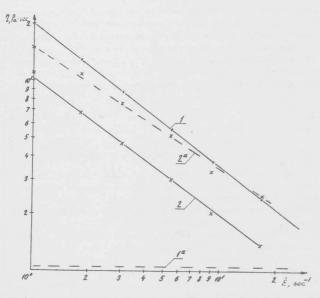
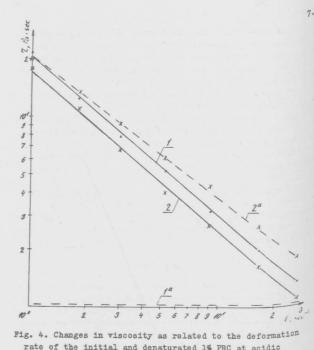


Fig. 3. Changes in viscosity as related to the deformation rate of the initial and denaturated 1% FRC at acidic and alkaline pH's at 20°C

- initial FRC - pH = 3.10 - pH = 3.10
- 1^a denaturated FRC
- 2 initial FRC

1

2^a - denaturated FRC and renaturated for 1.0; 1.5; 3.0; 5.0 and 20 hrs



$l = initial PRC$ $-pH = 3.10$ l^{a} - denaturated PRC $-pH = 3.10$ $2 = initial PRC$ $-pH = 8.30$ 2^{a} - denaturated PRC and renaturated	and alkaline pH at 20°C	110 90 901010
1^{a} - denaturated PRC ${DH} = 3.10$ 2 - initial PRC ${DH} = 8.30$ 2^{a} - denaturated PRC and renaturated	The second se	- pH = 3.10
2ª- denaturated PRC and renaturated	1 ^a - denaturated PRC	- pH = 3.10
2 ⁸ - denaturated PRC and renaturated for 20 hrs - pH = 8.30	2 - initial PRC	- pH = 8.30
for 20 hrs - pH = 8.30	2ª- denaturated PRC and renaturated	
	for 20 hrs	- pH = 8.30

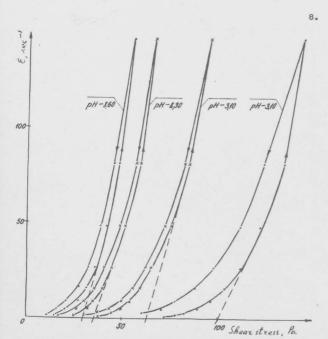


Fig. 5. Hysteresis loop for 0.8% PRC and 1.3% FRC in acidic and alkaline fields at 20°C

Acidic and alkaline PRC and FRC were heated in a thermostat at 40°C for 30 min. for "mild" heat denaturation to 00" cur with the destruction of the links stabilizing the native structure of collagen. After heating, denaturated PRC and FRC were cooled in a thermostat for 30-40 min. down to 20°C. The viscosity was measured just after heating and after cooling with the intervals of 1; 1.5; 3; 5 and 20 hrs. In acidic densturated FRC and PRC, the viscosity was noted to decrease near ly down to that of the solvent (Curves la in Figs. 3 and 4) and then to remain unlike and the solvent (Curves la in Figs. 3 and 4) and then to remain unaltered under shear stresses. In alkaline der naturated FRC and FRC just after heating the viscosity fell sharply, but afterwards it increased significantly (for 1.5 br in case of FRC and for 20 hrs in case of FRC), somewhat exceeded that of the initial solution in case of FRC), somewhat exceeded that of the initial solubilization products and became stable at this level (Curves 2a in Figs. 3 and 4). This process is; evidently, connected with the formation of a more perfect structure during the second ture during the renaturation of "mild"-denaturation products.

From the initial and denaturated soldio and alkaline pro ducts of collagen solubilization, films were cast on a poly the lene backing and air-dried. The welding temperature of these films was recovered films was measured, it being 43-47°C for the films from the int tial acidio and alkalter of the films from the tial acidic and alkaline solubilization products and 30-31°C for the films from denaturated alkaline products. Films from denaturated acidic products were dissolved in water during swelling without heating.

These results, as well as the data on viscosity, prove a significant effect of pH upon denaturation temperature and structure-mechanical properties of denaturated solubilization products.

For structured systems, with collagen solubilization products belonging to them, it was of interest to follow changes in the relative ultimate shear with pH (Fig. 5). For FRC with pH=3.10. this value above the pH=3.10, this value, characterizing structure strength, 15 equal to 100 Pa and for FRC with pH=8.60 it equals 30 Pa, i.e.

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to the acidic field is more than 3 times as high. This proves Ale acidic field is more than y they do a first in the acidic field is the structural network of FRC in the acidic Medium. As for PRC, this phenomenon is also observed but is less Pronounced (at pH=3.10 this value is equal to 63 Pa and at PH=8.30 - to 37 Pa).

The effect of shear stresses was studied on 1.3% FRC at Mag.10 and 8.60 and on 0.8% PRC at pH=3.10 and 8.30 as with the and 8.60 and on 0.0% for at particular of the viscosimeter slipped relative to the test solution at great deformation rates.

The effect of shift stresses upon FRC at pH=3.10 was found The effect of shirt stresses upon the viscosity and well-pronoun-result in a sharp change of the viscosity and well-pronoun-^{teg} hysteresis, this indicating the breakage of the structural Weteresis, this indicating the stating were smaller for he work. In the alkaline field these effects were smaller for The due to the development of a stronger structure. In case of Fac both in the acidic and alkaline fields, hysteresis was less pronounced.

It is known (8) that mechanical treatment is of a consi-^{4e}rable effect upon the viscosity of structured systems. With bis in view, we investigated a similar effect upon PRC and No in view, we investigated a similar pH value, viz., J. 10, Mechanical mixing was performed in a laboratory homogenither for 10 min. at 1,000 rpm. The data in the Table show that ¹⁰r 10 min. at 1,000 rpm. The data in the factor of the second solution viscosity rises greatly with PRC and FRC concentration of the strength the increasing up to 1.5%, this being due to structure strengthe-

Hydrated protein systems have a structural network which Bydrated protein systems have a structure attribute allo-to only very strong, but elastic and resident, this allo-Wing the above systems to damp the stresses from outside. Theretore, a mechanically destroyed hydrated protein system is able to recombine comparatively fast, this resulting in a new and Sometimes, a stronger structure.

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much less pronounced. 3. Heat denaturation of PRC and FRC solutions with pH = J. 10. Heat denaturation of PRC and rAC solutions whereas a sibullar treatment at pH = 8.30 or 8.60 somewhat increases the Viscosity of renaturated products as compared to the initial

out of renaturated products as compared to the formation of a more perfect structure from the product of the formation of a more perfect structure from the products of "mild" denaturation. 4. Mechanical treatments destroy considerably weakly-deve-

⁴. Mechanical treatments destroy constants ped aystems with a low protein concentration (up to 0.8%); Sime similar treatments of more concentrated systems (up to 1.5%) tens due to their high strength and elastic and resilient properties.

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Table

Changes in PRC and FRC viscosity at the deformation rate (È) of 1 sec.

Product	Concentra- tion, %		Viscosity after mixing, Pa•sec.	
FRC	0.8	13.60	3.93	71
PRC	0.8	26.90	9.40	65
FRC	1.5	45.50	36.00	21
PRC	1.5	112.00	103.5	7.5

Obviously, under the mechanical destruction of collagen solubilization products of a higher concentration, structure reconstitution is more vivid; post shear stresses the viscosity of concentrated systems during mixing decreases insignificantly. In systems of a low protein concentration, mechanical treatment causes great destruction of a poorly-developed structure and reduces its viscosity by nearly three times.

Conclusions

The results obtained allowed to draw the following conclusions:

1. Viscosity relation to PRC or FRC concentration in the range tested (0.8 to 2%) is of a linear character. FRC and PRC viscosity rises with protein level, this, probably, being due to the development of a stronger structural "carcass" in concentrated systems.

2. Changes in FRC pH from 3.10 up to 8.60 decrease the viscosity of the system by nearly two times, while similar changes in PRC pH cause only a slight fall in the initial value. Shear stresses alter sharply the viscosity of FRC having the pH of 3.10 with well-pronounced hysteresis. At pH = 8.60 for FRC and at pH either 3.10 or 8.30 for PRC, hysteresis is

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