# REACTIVITY OF HAEMATIN IN PROCESSED MEATS

D.A. LEDWARD, WENDY J. SCHOFIELD, T. HAZEL and R.J. NEALE University of Nottingham, Nottingham, England.

## INTRODUCTION

THE REACTIONS undergone by the haemoproteins of meat (mainly myoglobin) during storage, heating and curing have been undergone by the haemoproteins of meat (mainly myoglobin) during storage, heating and curing have been undergone by the haemoproteins of meat (mainly myoglobin) during storage, heating and curing have been undergone by the haemoproteins of meat (mainly myoglobin) during storage, heating and curing have been undergone by the haemoproteins of meat (mainly myoglobin) during storage, heating and curing have been undergone by the haemoproteins of meat (mainly myoglobin) during storage, heating and curing have been have been undergone by the haemoproteins of meat (mainly myoglobin) during storage, heating and curing have been have been undergone by the haemoproteins of meat (mainly myoglobin) during storage, heating and curing have been have bee have been extensively researched and comprehensively documented. However there is increasing interest in the development development and utilisation of freeze-dried and intermediate(I.M.) meats and although it is well established that the c that the freshly prepared products have the pigments in the same form as their full moisture counterparts while freshly prepared products have the pigments in the same form as their full moisture counterparts. Some we (a) the freshly prepared products have the pigments in the same form as their full moisture counterparts (Lawrie, 1979, Obanu and Ledward, 1975) there is some doubt as to their behaviour during storage. Some workers that the the terms of the pigments degrade to bile pigments (Lawrie, 1979) while Obanu and claim that the haemoproteins of freeze-dried meats degrade to bile pigments (Lawrie, 1979) while Obanu and Ledward (1975) there is some doubt as to their behaviour during storage. Ledward (1975) suggest that in I.M. meats the haematin remains intact, but firmly bound to the meat matrix.

In view of the importance of the haemoproteins and their environment, to both the colour and nutritional subject of the importance of the haemoproteins and their stability in dried and I.M. beef was undertaken.  $q_{uality}^{vLew}$  of the importance of the haemoproteins and their environment, to both the colour one standing of meat (Hazell et al, 1978) a study of their stability in dried and I.M. beef was undertaken.

MATERIALS AND METHODS

A BEEP 1.dorsi muscle, freed of visible fat and connective tissue, was cut in cubes (M1cm<sup>3</sup>) and samples either (a) freeze dried  $r_{c}$  (b) freeze dried (b) (c) Cooked in boiling water for 1 hour and freeze dried (c) cooked in boiling water ard processing in water at 260

 $v_{130}$  canned by hot filling and processing in watch to the processing match at the procesing match at the procesing match (a) were opened and the cubes dabbed with absorbent tissues to remove surplus fluid.

 $s_{anples}$  (a), (b) and (e) were packed in contact with air in PVDC Cryovac and together with samples (c) and (d)  $s_{anples}$  stored to  $s_{anples}$  (b) and (c) were packed in contact with air in PVDC Cryovac and together with samples (c) and (d)  $w_{ere}^{pues}$  (a), (b) and (e) were stored at -12 and +38°C. Spectral Analysis

The reflectance spectra of intact meat pieces was recorded against MgO using a Perkin-Elmer Model 124 spectro-Motometer fitte spectra of intact meat pieces was recorded against MgO using a Perkin-Elmer Model 124 spectro-set one-ter fitte spectra of intact meat pieces was recorded against MgO using a Perkin-Elmer Model 124 spectro-ter fitte spectra of intact meat pieces was recorded against MgO using a Perkin-Elmer Model 124 spectro-ter fitte spectra of intact meat pieces was recorded against MgO using a Perkin-Elmer Model 124 spectro-ter fitte spectra of intact meat pieces was recorded against MgO using a Perkin-Elmer Model 124 spectro-ter fitte spectra of intact meat pieces was recorded against MgO using a Perkin-Elmer Model 124 spectro-ter fitte spectra of intact meat pieces was recorded against MgO using a Perkin-Elmer Model 124 spectro-ter fitte spectra of intact meat pieces was recorded against MgO using a perkin-Elmer Model 124 spectro-ter fitte spectra of intact meat pieces was recorded against MgO using a perkin-Elmer Model 124 spectro-ter fitte spectra of intact meat pieces was recorded against MgO using a perkin-Elmer Model 124 spectro-ter fitte spectra of intact meat pieces was recorded against MgO using a perkin-Elmer Model 124 spectro-ter fitte spectra of intact meat pieces was recorded against MgO using a perkin-Elmer Model 124 spectro-ter fitte spectra of intact meat pieces was recorded against MgO using a perkin-Elmer Model 124 spectro-ter fitte spectra of intact meat pieces was recorded against MgO using a perkin-Elmer Model 124 spectro-ter fitte spectra of intact meat pieces was recorded against MgO using a perkin-Elmer Model 124 spectro-ter fitte spectra of ter fitte spect photometer fitted with a reflectance attachment. The range scanned was 700 to 370nm. The same instrument, set up in the Set up in the transmission mode, was used to record the absorption spectra of solutions at a path length of 1cm.

Solubility of Iron and Porphyrin

Although water and dilute salt solutions will readily extract the haemoproteins of raw meat in cooked meats he precipitate and dilute salt solutions can only be solubilised with difficulty (Ledward, 1971). the precipitated cooked meat haemoproteins can only be solubilised with difficulty (Ledward, 1971).  $\beta_{\rm reliminary}$  and dilute salt solutions will readily extract the haemoproteins or raw meat in the precipitated cooked meat haemoproteins can only be solubilised with difficulty (Ledward, 1971).

<sup>We</sup> precipitated cooked meat haemoproteins can only be solubilised with difficulty (Ledward, 1971). <sup>We</sup> uninary experiments using 40% pyridine, 6M area, 3% sodium dodecyl sulphate plus 1%  $\beta$ -mercaptoethanol, <sup>We</sup>  $\beta_{r_{0m}}$  HCl and were the most effective in extracting haematin  $0.41 m_{1}$  Putated cooked meat haemoproteins can only be accounted by a solution of the property of the pro from cooked meat.

Preliminary work also indicated that the extraction itself could lead to destruction of haematin from canned genty, M. Meator and Sector and the sector of and I.M. Meats. The most successful technique was to freeze dry the meat just prior to analysis followed by M. Meats. The most successful technique was to freeze dry the meat just prior to analysis followed by M. M. Powderin. The most successful technique was to freeze dry the appropriate solvent at 2°C for 48 hours. Sently work also indicated that the extraction reserve dry the meat just prior to analysis forfore a sently powdering in a pestle and mortar and then treatment with the appropriate solvent at 2°C for 48 hours. Cases the intercolvent was about 1:10. In all powdering in a pestle and mortar and then treatment cases the ratio of protein: solvent was about 1:10.

<sup>ron</sup> <sup>concentrations</sup> were determined by atomic absorption spectrophotometry (EEL Model 240) and porphyrin <sup>concentrations</sup> were determined by atomic absorption spectrophotometry (EEL Model 240) and porphyrin Concentrations were determined by atomic absorption.

Porphyrin concentrations of the solution were calculated as a percentage of that extracted from the raw meat Material and the solution of the iron content of the ashed freeze dried starting <sup>Nucphyrin</sup> concentrations of the solution were calculated as a percentage of that extracted from the form <sup>Naterial</sup> and iron concentration as a percentage of the iron content of the ashed freeze dried starting

RESULTS

NonAL iron content of the beef used was 7.89 mg/100g dry matter with a standard deviation of 0.08 on 12 samples iron on the content of the beef used was 7.89 mg/100g dry matter with a standard deviation of 0.08 on 12 samples <sup>4AL</sup> iron content of the beef used was 7.89 mg/100g dry matter with a standard deviation of 0.08 on 12 samples tron content of the beef used was 7.89 mg/100g dry matter with a standard deviation of 0.08 on 12 samples tron content of the beef used was 7.89 mg/100g dry matter with a standard deviation of 0.08 on 12 samples  $t_{ton}$  content of the beef used was 7.89 mg/100g dry matter. Due to the presence of glycerol and the change significantly during processing and storage. Due to the presence of glycerol and the content of the I.M. meat was 3.90  $\pm$  0.13 mg/100g dry matter which remained constant during storage. Bifect of Storage at 38°C on Reflectance Spectra

<sup>of Storage</sup> at 38<sup>o</sup>C on Reflectance Spectra <sup>In Figure 1</sup>the reflectance spectra of freshly prepared freeze dried raw beef is shown together with the spectra

of the samples stored in air at 38°C for both 4 weeks and 20 weeks. It is seen that the characteristic spectra of the fresh meat (predominantly oxymyoglobin) is lost during storage. During storage of cooked freeze dried meat under the same conditions the characteristic cooked meat haemoprotein spectra was lost. In agreement with the earlier work of Obanu and Ledward (1975) the I.M. meat also rapidly lost its haemoprotein characteristics during aerobic storage at 38°C (Fig. 2). However the anaerobically stored meat samples (both canned and I.M.) showed no loss of haemoprotein character during 20 weeks storage at 38°C (Fig. 2). In I.M. meat the loss of haemoprotein character appears to be rapider in heat treated samples (May, 1977).

Solubility in Acid and Alkali of Porphyrin and Iron in the Meats.

In all cases the majority of the iron (> 80 %) in the meats was extracted, in both the freshly prepared and stored meats, on treatment with either 0.1N HCl or N.NaOH. However during air storage the amount of porphyrin solubilised decreased.

The results for the alkali extraction of cooked and raw freeze dried beef are shown in Fig. 3 where it is seen that at  $-12^{\circ}$ C the raw pigment appears to be more stable than the cooked.



### 67

With the acid extraction the porphyrin is less stable as, although all of it can be extracted in the freshly prepared of the samples was extracted from the The acid extraction the porphyrin is less stable as, although all of it can be extracted in the frequency prepared freeze dried meats (both raw and cooked) only about 5% (mean of 4 samples) was extracted from the after 5 weeks at  $-12^{\circ}$ C. From the raw sample 30% (mean of two samples) was extracted by acid 38°C no Porphyrin, from either the raw or cooked freeze dried meat, could be extracted by acid (4 samples) and ysed).

Spectra of the solutions obtained verified these observations as marked haematin character (Lemberg and Legge, 1949) was <sup>1949</sup>, <sup>Was</sup> seen in the acid and alkaline solutions from the freshly processed meats but significantly less, or <sup>None</sup> at 22 None at all, in the stored freeze dried meats.

Even with the mild extraction procedure used in the present work it proved impossible to solubilise all the pophyrin the mild extraction procedure used in the present work it proved impossible to solubilise all the solution of 17,0, porphyrin from even freshly prepared I.M. meat. Less than 30% was extracted with N NaOH (values of 17,0, and 30 h.  $\frac{3}{4}$   $\frac{3}{30}$  being found for 4 samples) even though most of the iron was soluble (mean = 88 % on 4 samples). In  $0, \frac{and}{30}$  being found for 4 samples) even though most of the iron was soluble (mean = 00 × 0.1 +  $10^{-10}$  M HCl extracts it was not possible to detect any porphyrin in 4 samples although the iron solubility was  $10^{-10}$  m  $(30^{-10})$  high (> 90 %).

These degrees of extraction were apparently independent of storage time or temperature for the anaerobically stored same of extraction were apparently independent of storage time or temperature for the anaerobically <sup>Weske</sup> degrees of extraction were apparently independent of storage time or temperature for the anatomore stored samples but for the I.M. samples stored in air no porphyrin was extracted after 20 weeks at -12°C or <sup>Wesks</sup> at 38°C. Unfortunately no analyses were made at shorter time intervals.

In Canned meat, removed to air storage after processing and stored at  $-12^{\circ}$ C for 20 weeks about 50% (mean of 2 determinate), removed to air storage after processing and stored at  $-12^{\circ}$ C for 20 weeks about 50% (mean of 2 determinations) of the porphyrin could be extracted in NaOH compared to over 80% at zero time.

In all solutions from the I.M. and canned meats the presence or absence of haematin compounds was supported by the absence of haematin compounds was supported to the absence of haematin compounds are supported to the absence of haematin compounds was supported to the absence of haematin compounds was supported by the absence of haematin compounds was supported

 $b_{y}^{4l}$  solutions from the I.M. and canned meats the presence or absence of haematin compounds was support the absorption spectra; solutions giving positive porphyrin tests always displaying maxima in the Soret region.

### DISCUSSION

THE RESULTS reported above strongly suggest that during air storage the haematin pigments of freeze dried, 1, and a reported above strongly suggest that during air storage the haematin pigments of freeze dried, 1975). Ho I'M RESULTS reported above strongly suggest that during air storage the haematin pigments of freeze during in the and cooked meat are degraded to non-haematin compounds. It is possible to explain the changes observed even the reflectance spectra as being due to the formation of 'free' haematin (Obanu and Ledward, 1975). However, the above the a ever the reflectance spectra as being due to the formation of 'free' haematin (Obanu and Deaward, 'the absence of porphyrin in the alkali and acid extracts from the stored meats argue strongly against this as have the of porphyrin in the alkali in these solvents (Lemberg and Legge, 1949). this as haematin complexes should be stable in these solvents (Lemberg and Legge, 1949).

As the oxidation takes place in well heated meat and at sub-zero temperatures it is unlikely to be caused by either enzymes or bacteria. This non-enzyme oxidation of the haematin is apparently more rapid in cooked to react to react the rest of the reactivities and at elevated temperatures. Compared to raw meats, at I.M. water activities and at elevated temperatures.

The I.M. beef used in the present study was infused with glycerol which often contains certain reducible provide a study was infused with glycerol which often contains certain reducible provide a study was infused with glycerol which often contains certain reducible provide a study was infused with glycerol which often contains certain reducible the study was infused with glycerol which often contains certain reducible the study was infused with glycerol which often contains certain reducible the study was infused with glycerol which often contains certain reducible the study was infused with glycerol which often contains certain reducible the study was infused with glycerol which often contains certain reducible the study was infused with glycerol which often contains certain reducible the study was infused with glycerol which often contains certain reducible the study was infused with glycerol which often contains certain reducible the study was infused with glycerol which often contains certain reducible the study was infused with glycerol which often contains certain reducible the study was impurities and these impurities have been shown to totally destroy the haematin of myoglobin even when the it was and these impurities have been shown to totally destroy the haematin of myoglobin even when the it was and these impurities have been shown to totally destroy the haematin of myoglobin even when the it was and these impurities have been shown to totally destroy the haematin of myoglobin even when the it was a shown to totally destroy the haematin of myoglobin even when the it was a shown to totally destroy the haematin of myoglobin even when the it was a shown to totally destroy the haematin of myoglobin even when the it was a shown to totally destroy the haematin of myoglobin even when the it was a shown to totally destroy the haematin of myoglobin even when the it was a shown to totally destroy the haematin of myoglobin even when the it was a shown to totally destroy the haematin of myoglobin even when the it was a shown to totally destroy the haematin of myoglobin even when the it was a shown to totally destroy the haematin of myoglobin even when the it was a shown to totally destroy the haematin of myoglobin even when the it was a shown to totally destroy the haematin of myoglobin even when the it was a shown to totally destroy the haematin of myoglobin even when the it was a shown to totally destroy the haematin of myoglobin even when the it was a shown to totally destroy the haematin of myoglobin even when the it was a shown to totally destroy the haematin of myoglobin even when the it was a shown to totally destroy the haematin of myoglobin even when the it was a shown to totally destroy the haematin of myoglobin even when the it was a shown to totally destroy the haematin of myoglobin even when the it was a shown to totally destroy the haematin of myoglobin even when the shown to totally destroy the haematin even when the shown to totally destroy the haematin even when the shown to totally destroy the haematin even when the shown to totally destroy the haematin even when the shown to totally protein was not denatured (Bello and Bello, 1976). However freeze dried meat does not contain glycerol but the does have been shown to totally destroy the haematin of myoground even with the second does not contain glycerol but the does have been shown to totally destroy the haematin of myoground even with the second does not contain glycerol but the does have been shown to totally destroy the haematin of myoground even with the second does not contain glycerol but the does have been shown to totally destroy the haematin of myoground even but the does have been shown to totally destroy the haematin of myoground even but the does have been shown to totally destroy the haematin of myoground even but the does have been shown to totally destroy the haematin of myoground even but the does have been shown to totally destroy the haematin of myoground even but the does have been shown to totally destroy the haematin of myoground even but the does have been shown to totally destroy the haematin of myoground even but the does have been shown to totally destroy the haematin of myoground even but the does have been shown to totally destroy the haematin of myoground even but the does have been shown to totally destroy the haematin of myoground even but the does have been shown to totally destroy the haematin of myoground even but the does have been shown to totally destroy the haematin of myoground even but the does have been shown to totally destroy the haematin of myoground even but the does have been shown to totally destroy the haematin of myoground even but the does have been shown to totally Subject the state of the second state of the sta <sup>during</sup> the oxidation of muscle lipids can destroy the haematin ring in haemoproteins (Roubal and Tappel, 1966). Support for the hypothysis that the intermediates formed during the oxidation of muscles lipids are lipid oxidation in the hypothysis that the intermediates formed during the oxidation of muscles lipids are lipid oxidation in the haematin destruction these processed meats is given by the following observations; (i) althe oxidation is most realized in the LaM, moisture range i.e. at a 's at which haematin destruction is most realized whole blood <sup>Sponsible</sup> for the hypothysis that the intermediates formed during is given by the following observations, (1) <sup>lipid</sup> oxidation is maximal in the I.M. moisture range i.e. at a 's at which haematin destruction is most rapid <sup>shows</sup> little l <sup>lipid</sup> or course glycerol impurities may also contribute in these samples (ii) freeze dried whole blood <sup>lipid</sup> oxidation is maximal in the restore during air storage (Hazell, 1979) and (iii) the rate of destruction is <sup>lipid</sup> oxidation is known to be <sup>th</sup>ough of course glycerol impurities may also contribute in these samples (ii) freeze dried whole second free the second structure in these samples (iii) freeze dried whole second free the second structure in the second second structure is the second structure in the second secon

And the second s More rapid in cooked than fresh meat. Portunately this loss of haematin does not appear to affect the high nutritional availability of the iron <sup>Sociated</sup> with the second second

<sup>ACUNAtely</sup> this loss of haematin does not appear to a <sup>ACSBOCLATED</sup> with meat (Obanu et al, 1976, Hazel, 1979). REFERENCES Rello, J. and Bello, H.R. 1976. Arch. Biochem. Biophys. <u>172</u>, 608. Rello, T. 1979. Ph.D. Thesis. University of Nottingham. Ledwrie, R.A. 1970. Mart Signece 3rd edition. Pergamon Press. Oxf Wel, T. and Bello, H.R. 1976. Arch. Diversity of Nottingham. Ledward, R.A. 1979. Ph.D. Thesis. University of Nottingham. Ledward, D.A. 1979. Meat Sicnece 3rd edition. Pergamon Press. Oxford. Oban, 2.A. 1971. J. Food Sci. <u>36</u>, 883. Oban, 2.A. and Ledward, D.A. 1975. J. Food Technol. <u>10</u>, 675. Oban, 2.A. Lodward, D.A. and Lawrie, R.A. 1975. J. Food Technol. <u>10</u>

Vanu ', U.A. 1971. J. Food Sci. <u>36</u>, 805. Danu, Z.A. and Ledward, D.A. 1975. J. Food Technol. <u>10</u>, 675. Danu ', Z.A.', Ledward, D.A. and Lawrie, R.A. 1975. J. Food Technol. <u>10</u>, 657. Lemp, Z.A. Di Varle R.J., Ledward, D.A. and Lawrie, R.A. J. Food Man. Di Varle R.J., Ledward, D.A. and Lawrie, R.A. J. Food Man. Di Varle R.J., Ledward, D.A. and Lawrie, R.A. J. Food Man. Di Varle R.J., Ledward, D.A. and Lawrie, R.A. J. Food Man. Di Varle R.J., Ledward, D.A. and Lawrie, R.A. J. Food Man. Di Varle R.J., Ledward, D.A. and Lawrie, R.A. J. Food Man. Di Varle R.J. Ledward, D.A. and Lawrie, R.A. J. Food Man. Di Varle R.J. Ledward, D.A. and Lawrie, R.A. J. Food Man. Di Varle R.J. Ledward, D.A. and Lawrie, R.J. Ledward, D.A. and Lawrie, R.J. Man. J. Food Man. Di Varle R.J. Ledward, D.A. and Lawrie, R.J. Ledward, D.A. and Ledward, D.A. Manu, Z.A. and Ledward, D.A. 1975. J. Food Technol. 10, 657. Danu, Z.A. Ledward, D.A. and Lawrie, R.A. 1975. J. Food Technol. 10, 657. Mayorg, R. A., Biggin, R.J., Neale, R.J., Ledward, D.A. and Lawrie, R.A. J. Food Technol. 11, 883. Mayorg, R. and Lenu, 2.A., Ledward, D.A. and Lawrie, R.A. 1975. 5. 1000 Lenuberg, Z.A., Biggin, R.J., Neale, R.J., Ledward, D.A. and Lawrie, R.A. J. Food Technol. 11, 0000 May, J.R. and Legge, J.W. 1949. Haematin compounds and bile pigments. Interscience. New York. Poubal. D. B.S. Hegge, J.W. 1949. Haematin compounds and bile pigments. Interscience. New York. May, J.D. B. Sc. Hons. Dissertation. University of Nottingham. Hay, J. D. B. Sc. Hons. Dissertation. University of Nottingham. Hay, J. D. B. Sc. Hons. Dissertation. University of Nottingham. Hay, J. Nutr. <sup>YY</sup>, J.D. <sup>R.</sup> and Legge, J.W. 1949. Indematic Roubal D. B.Sc. Hons. Dissertation. University of Nottingham. <sup>Hazel</sup>, W.T.R. and Tappel, A.L. (1966) Arch. Biochem. Biophys. <u>113</u>, 5. <sup>T.</sup> Ledword Tappel, R.J. 1978. Brit. J. Nutr. <u>39</u>, 631

Mubal, W.T.R. and Tappel, A.L. (1966) Arch. Biochem. Biophys. <u>115</u>, 631. <sup>Hazel</sup>, T., Ledward, D.A. and Neale, R.J. 1978. Brit. J. Nutr. <u>39</u>, 631.