INTERFACIAL AND EMULSIFYING PROPERTIES OF BLOOD PLASMA PROTEINS.

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RODUCTION

<sup>bog</sup> plasma, derived from slaughterhouse blood by centrifuging off the red blood cells, <sup>bog</sup>tl, No. Plasma, derived from slaughterhouse blood by centringing off is distributed either trocen used as an additive in different sausage formulations. It is distributed either trozen or spraydried. The functionality of the blood plasma proteins in sausage pro-blasma been found to be good. For example a certain exchange of the meat proteins for blasma proteins gives a product, which retain its fat and water binding properties plasma proteins gives a product, which retain its fat and water binding properties,  $l_{0hg}$  as the temperature in the product is raised to  $76^{\circ}C$ .

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<sup>18</sup> Well established that the plasma proteins have very good gelling properties (1-6), the Optimum in waterbinding properties of these gels is obtained at 74-80°C (3). Inthe optimum in waterbinding properties of these gels is obtained at ratio of the per-stigations of the emulsifying properties of blood plasma proteins have also been per-The display of the emulsifying properties of blood plasma proteins have also been per-emulsifiers. However, the elucidation of proteins as emulsifiers is not straight for-tring several contributing factors during the emulsification process can give con-time results, if not kept under control. Firstly, the emulsifying apparatus, intensity emulsions formed (11), which in turn influences the emulsion stability. Therefore the two cation process has to be accurately standardized in order to make any comparison <sup>shull are dominating parameters in string process the emulsion stability. Increasion is a stability of the </sup> Ween proteins regarding their emulsifying properties. Secondly, the amount of protein and proteins regarding their emulsifying load) is largely determined by the type of protein Ween Proteins regarding their emulsifying properties. Secondly, the amount of protein founding the fat globule (protein load) is largely determined by the type of protein the maximus and the protein/fat surface area ratio (11). This latter fact has to be considered have and sifying capacity measurements as used by most authors. In these measurements here are a mount of fat emulsified by a protein dispersion of a certain protein content the emulsifying constantly stirred, until the emulsion inverts into a water-in-oil emulsion. here been found that the solubility of the protein correlates very well has being constantly stirred, until the emulsion inverts incompany well been found that the solubility of the protein correlates very well at the the found that the solubility of the protein correlates very well the found that the solubility of the protein correlates very well is to be expected, as mainly the soluble part of the protein acts as an emulsifier. influence of the insoluble part of the protein. It still turned out to be the most hot protein that had the greatest ability to emulsify fat (8). From these findings one lity to emulsify fat because the more soluble protein has a higher protein/fat surface oluble. Not Protein that had the greatest ability deduce that the less soluble protein contains soluble protein with a formation of the solution of th  $r_{atio}^{tO}$  emulsify fat, because the more soluble protein nas a higher protein membrane during the measurements. This results mostly in a thicker protein membrane if the molecular weight of the tatio emulsify fat, because the more solution mostly in a thicker protein memorance build the during the measurements. This results mostly in a thicker protein memorance build the fat droplets for the more soluble proteins, if the molecular weight of the there is polydisperse (12). This is the case for most protein products. The thicker the there acial is that the emulsion will be stable. Weing is polydisperse (12). This is the case for most protein products. The thread facial layer of the protein the more probable it is that the emulsion will be stable.

<sup>Autal</sup> layer of the protein the more proteins <sup>Autal</sup> layer of the protein the more proteins are very much dependant on the <sup>Autal</sup> sifying conditions and on the protein concentration. Therefore these factors have to <sup>Autal</sup> controlled a second to this, a procedure has been worked out, where the emulsi-autal a whey protein concen-The emulsifying properties of proteins are interface as compared with the emulsifying properties of blood plasma proteins and to investigate the interthe behaviour of blood plasma proteins studied before (19). behaviour of blood plasma proteins at the soybean oil/water interface as compared

AND METHODS

AND METHODS Ren Serum albumin (BSA) from Sigma Chemical Co. was used. Delocation of the serum albumin (BSA) from Sigma Chemical Co. was used. Serum albumin (BSA) from Sigma Chemical Co. was used. (1) Jood plasma from Ellco Protein, Kävlinge, Sweden was used. Analysis: protein dried 1, wt%, fat 0.5 wt%, salt 1.5 wt%.

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<sup>e chloride</sup> solution at pH 7, denoted as (0-7) and (0,2-7) for the soy protein <sup>bource</sup>, protein content and solubility in (0-7) and (0,2-7) for the soy protein <sup>bource</sup>, protein content and solubility in (0-7) and (0,2-7) for the soy protein <sup>bource</sup>, protein content and solubility in (0-7) and (0,2-7) for the soy protein <sup>bource</sup>, protein content and solubility in (0-7) and (0,2-7) for the soy protein <sup>bource</sup>, protein content and solubility in (0-7) and (0,2-7) for the soy protein <sup>bource</sup>, protein content and solubility in (0-7) and (0,2-7) for the soy protein <sup>bource</sup>, protein content and solubility in (0-7) and (0,2-7) for the soy protein <sup>bource</sup>, protein content and solubility in (0-7) and (0,2-7) for the soy protein <sup>bource</sup>, protein content and solubility in (0-7) and (0,2-7) for the soy protein <sup>bource</sup>, protein content and solubility in (0-7) and (0,2-7) for the soy protein <sup>bource</sup>, protein content and solubility in (0-7) and (0,2-7) for the soy protein <sup>bource</sup>, protein content and solubility in (0-7) and (0,2-7) for the soy protein <sup>bource</sup>, protein content and solubility in (0-7) and (0,2-7) for the soy protein <sup>bource</sup>, protein content and solubility in (0-7) and (0,2-7) for the soy protein <sup>bource</sup>, protein content and solubility in (0-7) and (0,2-7) for the soy protein <sup>bource</sup>, protein content and solubility in (0-7) and (0,2-7) for the soy protein <sup>bource</sup>, protein content and solubility in (0-7) and (0,2-7) for the soy protein <sup>bource</sup>, protein content and solubility in (0-7) and (0,2-7) for the soy protein <sup>bource</sup>, protein content and solubility in (0-7) and (0,2-7) for the soy protein <sup>bource</sup>, protein content and solubility in (0-7) and (0,2-7) for the soy protein <sup>bource</sup>, protein content and solubility in (0-7) and (0,2-7) for the solution and (0,2-7) for the sol

The analysis is given elsewhere (11). <sup>the</sup>rcially available <u>soybean oil</u> (AB Karlshams Oljefabriker, Karlshamn, Sweden) was The analysis is given elsewhere (11).

The interfacial tension decay of the proteins at the soybean oil/water interface as a function of time was monitored with an apparatus based on the drop volume technique. A dest cription of the apparatus used (17) and the procedure worked out for measuring time-dependent interfacial tensions (18) have been been appared by the procedure worked out for measuring main time-dependent interfacial tensions (18) have been been appared by the procedure worked out for measuring main terms of the procedure worked out for measuring time-dependent interfacial tensions (18) have been been appared by the procedure worked out for measuring time-dependent interfacial tensions (18) have been appared by the procedure worked by the pr time-dependent interfacial tensions (18) have been given earlier. The temperature was  $m^{ain'}$  tained at 25+0.1°C.

Protein stabilized emulsions were made up of 40% (w/w) soybean oil and 60% (w/w) protein dispersion of varying protein content. A quantity of 50 gram was emulsified in a valve homogenizer or a sonifier incorporated into a recirculating system as previously described (21). The flow rate of the emulsion through the recirculating system was held constant at 250+25 ml min<sup>-1</sup>, and the temperature of the system was  $25+2^{\circ}\text{C}$  during and the temperature of the system was 25+2°C during processing. The power and energy input during the emulsification process have been measured (21). The emulsions were stored for 24 hr at 20<sup>o</sup>C and thereafter characterized in terms of fat particle size distribution and amount of protei protein adsorbed per unit area of fat surface  $(mg/m^2)$  (14). The interfacial area of the fat particles expressed in  $m^2$  fat/ml emulsion is derived from the particle size distribution according to calculations given elsewhere (22).

#### RESULTS AND DISCUSSION

### SDS-Polyacrylamide gel electrophoresis

Separation of the proteins in BSA, frozen blood plasma and spray dried blood plasma can be seen in figure 1. The major protein in blood plasma is BSA, and this is consistent with the approximate content of 55-64% of albumin in blood as given by Feeney (16). The protein bands in the spray dried plasma are not as distinct as for the frozen plasma, and the amount of BSA has also decreased. Evidently, some changes occur to the proteins during spray drying, which also shows up when comparing the interfacial behaviour of the two differently treated plasma p

treated plasma proteins (see below). Compared to BSA the two plasma protein products have a more polydisperse protein pattern, which is especially evident in the high molecular weight region. Besides albumin blood

also contains lipoprotein 4-14% and immunoglobulines (1-10%), wh molecular weight (16). which are proteins of high

As a contrast sodium caseinate and WPC, both from milk, have a narrow molecular weight pattern. The pattern of soy protein isolate is more polydisperse but not as much as for the blood plasma protein.

#### Interfacial behaviour

The interfacial tension-time  $(\gamma-t)$  dependence resulting from the adsorption of BSA, frozen and spray dried plasma is presented in figure 2. The initial bulk phase concentration of the proteins was  $10^{-2}$ % (w/w) based on the soluble content. They were dispersed in distilled water at pH 7 (0-7). In this concentration range the surface activity is highest for BSA, followed by frozen blood plasma and spray dried blood plasma, having the lowest interfacial depression effect. It is interesting to note that the difference in interfacial activity is greater between spray dried and frozen blood plasma than between frozen blood in the blood plasma contributes in this concentration range to a large extent to the surrace activity of blood plasma. Tybor et al. (9) have shown that the solubility of the plasma proteins is reduced by spray drying, and our results show that also the interfacial activity of the proteins is altered during the spray drying operation. plasma and BSA. The small difference between the

The concentration-dependence of the interfacial tension obtained at the soybean  $oil^{-water}$ interface for the spray dried blood plasma at (0-7) and (0.2.7) in the soybean at at interface for the spray dried blood plasma at (0-7) and (0.2-7), the soy protein at (0.2-7) and (0-7), the sodium caseinate at (0.2-7) and (0-7) and the WPC at (0.2-7) and (0-7) can be followed in figure 3 (19). In this figure the surface pressure

Figure 1. SDS-polyacrylamide gels showing the bands corresponding to 20, 40 and 80 µg of (from the left) BSA, frozen blood plasma and spraydrie plasma and spraydried blood plasma

30 (1-m

Nm

tension 20 Interfacial 10 50 40 30 10 20 TIME (minutes)

Figure 2. Time-dependence of interfacial tensions at the soybean oil-water interface for BSA (0), broydried blood plasma (**m**) and spraydried blood plasma (**x**) at (0-7) and a sybphase concentration of a subphase concentration of  $10^{-71}$  % (w/w)

%<sup>4An molecules can pack more densig at the second se</sup> Paring the concentration dependence of the erfacion the proteins studies and the proteins studies and the proteins studies are all the proteins are all terfacial tension for all the proteins studied before the stated. When salt is added, Following can be stated. When salt is added, sodim <sup>s-4</sup>Owing can be stated. When sale curve <sup>sodium</sup> caseinate and the WPC have a curve similar to each other giving a plateau tein no such independence of concentration type until tration of 1%. Furthermore baud Until a concentration of 1%. Furthermore,

hained after 40 minutes,  $\pi 40$  min is plotted Minst the initial subphase concentration. The Mace have a concentration of the fined as  $\gamma = \gamma_0 - \gamma$ , The initial subphase  $\gamma = \gamma_0 - \gamma_0^2$ , is defined as  $\gamma = \gamma_0 - \gamma_0^2$ , is the pressure,  $\pi$ , is defined as  $\gamma = \gamma_0 - \gamma_0^2$ , is the initial interfacial tension. In the figure it can be stated for all the initial interface over the with teins studied that the addition of 0.2 M NaCl Ans studied that the addition of other whole ances the interfacial activity over the whole central the plasma protein. Centration range. Although the plasma proteins tion 18% salt, which gives a salt concen-tion of 0.06 M in a 1% protein dispersion, ation Salt addition has a large influence on tr surface activity. When salt is added the trical double layer surrounding the protein loid is suppressed, and the electrostatic halt is suppressed, and the electrostate ad-thed molecules is reduced. Therefore the tein molecules is reduced. Therefore the terface

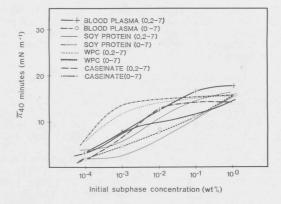


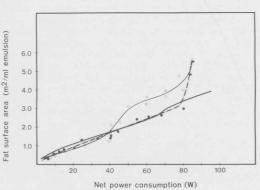
Figure 3. The surface pressure attained at the soybean oil-water interface after 40 minutes, 7740 minutes. as a function of the initial sub-phase concentration for all the proteins studied at different ionic strengths (19)

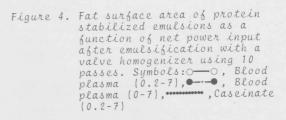
interfacial tension of 1%. Furthermore, interfacial tension decay caused by the milk teins is much larger (caseinate always being superior) compared to the other two eins at the lower concentrations, whereas the reverse is found at the highest concen-tions. This behaviour is especially pronounced for the blood plasma proteins, which opeater surface activity than the soy proteins in the concentration range studied. <sup>95</sup>. This behaviour is especially pronounced for the blood plasma proteins, <sup>9</sup>n<sub>cater</sub> surface activity than the soy proteins in the concentration range studied. <sup>9</sup>n<sub>ly</sub> properties such as hydrophobicity, flexibility and charge density of the <sup>10</sup>n<sub>s</sub>, but teins, the subjective such as hydrophobicity, flexibility and charge density of the subjective such as hydrophobicity, flexibility and charge density of the subjective such as hydrophobicity, flexibility and charge density of the subjective such as hydrophobicity, flexibility and charge density of the subjective such as hydrophobicity, flexibility and charge density of the subjective such as hydrophobicity, flexibility and charge density of the subjective such as hydrophobicity, flexibility and charge density of the subjective such as hydrophobicity, flexibility and charge density of the subjective such as hydrophobicity, flexibility and charge density of the subjective such as hydrophobicity, flexibility and charge density of the subjective such as hydrophobicity, flexibility and charge density of the subjective such as hydrophobicity, flexibility and charge density of the subjective such as hydrophobicity, flexibility and the hydrophobicity, flexibility and charge density of the subjective such as hydrophobicity, flexibility and the subjective such as hydrophobicity, flexibility, flexibi polymer/surface area ratio increases a kind of fractionation occurs, with the high <sup>yme</sup>r/surface area ratio increases a kind of fractionation occurs, when the solution. The <sup>electrophoresis</sup> runs have shown that both the blood plasma and the soy proteins have a be molecular weights on the surface and the the blood plasma and the soy proteins have a be molecular weights are shown that both the blood plasma and the soy proteins have a be molecular weights are shown that both the blood plasma and the soy proteins have a be molecular weights are shown that both the blood plasma and the soy proteins have a be molecular weights are shown that both the blood plasma are shown and the soy proteins are shown and the soy proteins have a be molecular weights are shown that both the blood plasma are shown and the soy proteins have a be molecular weights are shown that both the blood plasma are shown and the soy proteins have a be molecular weights are shown that both the blood plasma are shown are molecular weight and a wider distribution than the milk proteins. According to the outly Succular weight and a wider distribution than the mirk proteins. According outlined above this could explain the higher surface activity of the blood plasma sources sources. the soy proteins at the very high concencions

(large protein/surface area ratio). At lower wer concentrations the importance of phobicity and the flexibility of the increases in determining the interfacial wity, m. Y. This might explain the superior inter-activity of the caseinate and the WPC, having higher hydrophobicity and being more ble coming higher hydrophobicity and being more ordered exible compared to BSA, having a more ordered 468.  $\frac{v_{re}}{v}$  with 17 S-S-bridges and a helix content

# Usifying behaviour

Figure 4 the fat surface area of the emulsions with the fat surface area of the recirculating induce 4 the fat surface area of the emotion of the fat surface area of the recirculating with 10 passes through the recirculating ve are plot of the power input during are plotted versus power input during We homogenization. The initial protein content w) continous phase was in all cases 2.5% continues phase was in all cases 2.56 drift the proteins used as stabilizers were hate (0.227) mbc latter protein was chosen  $d_{ried}$  proteins used (0-7) and (0.2-7) and (0.2-7) and (0.2-7) and (0.2-7) metater protein was chosen  $d_{rieg}$  ample of non-flocculating emulsions,  $d_{ried}$  almost these increase of fat surface an almost linear increase of fat surface "ith Power input, when valve homogenized Ce area of (11, 14) has found that the fat Ce area of the stabilized emulsion, face The area of a protein stabilized emulsion, the area of a protein stabilized emulsion, ther in a sonifier or in a valve homo-loccul larger for flocculating than for the protect of th





Ner, is larger in a sonifier or in a valve nome-locculating emulsions. By microscopical examination it could be stated that floccu-on was observed around 40 W for blood plasma (0,2-7) and around 75 W for blood plasma is in Second to the two consumptions coincide well with a sudden in-tized emulsions. Otherwise the curves follow the more or less linear dependence with consumption. d8e

This further substantiates the findings by Tornberg (11) that the emulsifying conditions, such as emulsifying intensity and time, are dominating with regard to the final droplet size distribution, the choice of protein being of minor importance.

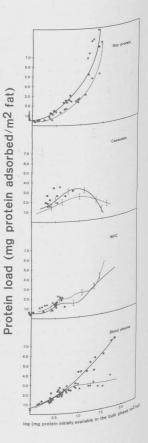
As has been pointed out throughout this article the protein/fat surface area ratio is a crucial factor in determining the amount of protein adsorbed on the fat droplets (protein load). In figures 5 and 6 protein load (mg adsorbed/ $m^2_{fat}$ ) and percentage protein adsorbed from the bulk phase is plotted as a function of mg protein ini-tially available in the bulk phase/ $m^2_{fat}$  (X) on a log scale. All the protein stabilized emulsions at (0-7) and (0,2-7) are included in the figures (13). The results obtained for the blood plasma and the soy protein (0,2-7) stabilized emulsion are plotted in the figures, whereas lines with error of limits are drawn for the other protein stabilized emulsions. Log X can be varied in two ways, either by having an initial constant protein concentration in the bulk phase and expanding the fat surface area by increased emulsifying intensity, or by varying the initial protein concentration and keeping the fat surface area more or less constant by the same emulsifying procedure. The former procedure has resulted in the lines with error of limits, whereas results obtained from the latter procedure are marked as points in the figures. The discrepancy between these two ways of making the emulsions will be dis-cussed more thoroughly elsewhere (13). The main conclusion that can be drawn from this investigation is that the probability increases for a higher protein load when varying the protein content, due to more coalescence occuring during emulsification.

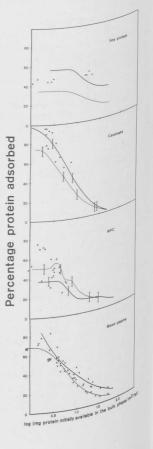
Figure 5. Protein load of all the protein stabilized emulsions at (0-7)  $(\bigstar -- \bigstar)$  and (0.2-7)  $(\bigstar -- \bigstar)$  as a function of log (mg protein initially available in the bulk phase/ $m^2$  fat)

If we compare the different proteins in figure 5 at high log X, i.e. at a high protein/fat surface area ratio, very different behaviour is observed. The soy proteins, especially in 0,2 M NaCl, where aggregation of the proteins is favoured, have a high protein load ( $\approx 10 \text{ mg/m}^2$ ), whereas for the caseinates it is as low as 0,5-1,5 mg/m<sup>2</sup>. Moreover, the caseinates do not show the typical curve form observed for the other protein stabilized emulsions, where protein load increases or levels out for higher log X. There is instead a maximum in protein load around log X 1,0. If we follow the theory of Cohen Stuart et equals to al. (12) it suggests that protein load should increase with higher protein/surface area ratios as there is a preference for the higher molecular weight part of the proteins to adsorb at the interface. This is followed by the whey proteins, the soy proteins and the blood plasma proteins but not by the caseinates. Evidently, the caseinates have no high molecular part to adsorb, which further substan-tiate the proposal made by Tornberg (22) that it is the monomeric casein, which is in equilibrium with the casein aggregate, that migrates to the interface. Blood plasma (0,2-7) has a high protein load at large log X in between the soy proteins and the WPC. For blood plasma dispersed in distilled water the high level of protein load is not achieved, but instead the amount adsorbed stays around 2,8  $\rm mg/m^2$  independent of log X.

The behaviour of the proteins at low log X can better be viewed in figure 6. There it can be seen that for the caseinates the percentage protein adsorbed increases with decreasing log X even to the lowest protein/surface area

> Figure 6. Percentage protein adsorbed from the bulk phase to the interface of all the protein stabilized emulsions at (0-7) (★---★) and (0.2-7) (●---●) as a function of log (mg protein initially available in the bulk phase/m<sup>2</sup> fat)





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<sup>tatio</sup>, whereas for the soy proteins it levels out already at a log X around 1,0. This  $g_{ives}$  in this low log X-region thicker protein membranes for the caseinate  $g_{ives}$  in this low log X-region thicker protein stabilized emulsions. At high Sives in this low log X-region thicker protein membranes for the caseInate stabilizedemulsions than for the soy protein stabilized emulsions. At high log X the reverse is found. The blood plasma proteins especially in (0,2-7) behaves more like the caseInates in the low log X-region. This is favourable, because the amount adsorbed for the blood plasma proteins is in this region higher than for the WPC and the soy proteins but less than for the caseInates. Thus, the blood plasma proteins seem to be good but less than for the caseinates. Thus, the blood plasma proteins seem to be good emulsifiers, anyhow when they are dispersed in 0,2 M NaCl solution, because comparatively thick protein membranes are achieved both at low and high protein/fat surface area ratios. REFERENCES 1. Hermanson, A.-M. and Tornberg, E. 1976. 22nd European Meeting of Meat Research Workers, Malmö, Sweden, I 1:3. 2. Hermansson, A.-M. 1978. 24th European Meeting of Meat Research Workers, Kulmbach, West Germany, H 1:3. 3. Hermansson, A.-M. Gel characteristics - Waterbinding properties of plasma protein gels. Submitted for publication in J. Sci. Fd Agric. 4. Fretheim, K. and Gumpen, S.A. 1978. 24th European Meeting of Meat Research Workers, Kulmbach, West Germany, H 9:3. S. Barper, J.P., Suter, D.A., Dill, C.W. and Jones, E.R. 1978. J. Food Sci. 43, 1204. Gumpen, S.A. and Fretheim, K. 1980. 26th European Meeting of Meat Research Workers, Colorado, USA, C-4. 7. Tybor, P.T., Dill, C.W. and Landmann, W.A. 1973. J. Food Sci. 38,4. 8. Satterlee, L.D., Free, B. and Levin, E. 1973. J. Food Sci. 38, 306. <sup>9</sup>. Tybor, P.T., Dill, C.W. and Landmann, W.A. 1975. J. Food Sci. 40, 155. <sup>10</sup>. Marshall, W.H., Dutson, T.R., Carpenter, Z.L. and Smith, G.C. 1975. J. Food Sci. 40, 896 11. Tornberg, E. 1980. J. Food Sci. 45, 1662. 12. Cohen Stuart, M.A., Scheutjens, M.H.M. and Fleer, G.J. 1980. J. Polymer Sci: Polymer Physics Edition, vol. 18, 559. <sup>13.</sup> Tornberg, E. and Pilman, E. To be published. 14. Tornberg, E. 1978. J. Sci. Fd Agric. 29, 867. <sup>15.</sup> Neville, M.D. 1971. J. Biol. Chem. 246, 20. <sup>16.</sup> Feeney, R.E. and Allison, R.G. Evolutionary biochemistry of proteins. Wiley Interscience, 1969 New York. <sup>17</sup>. <sup>Tornberg, E. J. Coll. Interface Sci. 1977, 60, 50.</sup> <sup>18.</sup> <sup>Tornberg, E. J. Coll. Interface Sci. 1978, 64, 391.</sup> <sup>19</sup>, <sup>Tornberg</sup>, E., Granfeldt, Y. and Håkansson, C., Submitted for publication in J. Sci. Fd Agri <sup>341</sup>C. <sup>341</sup>C. <sup>Mit</sup>chell, J., Irons, L. and Palmer, G.J. 1970. Biochem. Biophys Acta, 200, 138. <sup>2]</sup>. <sup>Tornberg</sup>, E. and Lundh, G. 1978. J. Food Sci. 43, 1553.

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