

INFLUENCE OF FREEZING ON TYROSINE PRECIPITATE IN SPANISH RAW CURED HAM

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INTRODUCTION

In Spanish raw cured hams, tyrosine precipitates can be found in two different forms: as crystals in the flesh and white film in cut surface (Silla et al., 1985, Arnau, et al., 1986).

Several theories have been elaborated in order to explain the origin of free tyrosine in the muscle:

Artioli (1952) found some correlation between blastomycetic flora and free tyrosine in the muscle. Comi et al., (1981, 1982, 1983) in Parma hams isolate yeast species of *Debaryomyces*, *Torulopsis* and *Trichosporon* that could synthesize tyrosine.

However proteolytic activity of cathepsins is the more feasible theory because the microbial counts in the muscle are too small to justify the proteolysis. Melo et al., (1974) observed a decrease in catheptic activity after an aging period of six months, and holding hams in frozen storage at -29°C for long periods were not detrimental to specific enzyme activity. Sárraga et al., (1988) found an increase of calpain activity that reaches its maximum two months after the beginning and remains constant until the end. Cathepsin L shows a constant activity during all the process and cathepsin D decreased progressively since the fourth month.

The main component of white film was tyrosine (Butz et al., 1974) and the relative proportion of phenylalanine is also increased (Arnau et al., 1987). Storage of ham slices at -18°C inhibits white film, but it appears when ham is thawed. White film decreases as the thickness of vacuum packaged ham slices diminishes, and does not

appear when cut surface is treated with high adherence products such as plastic solutions and gelatine (Arnau, et al., 1988), white film is chiefly found in muscles with the smaller percentage of salt in the beginning of the process that have the higher percentage of salt in the ham at the end of the process.

In Spain an important percentage of raw cured hams are manufactured from previously frozen hams. Thus the objective of these study was to analyze the influence of freezing on tyrosine precipitation.

MATERIAL AND METHODS

23 animals were sampled. In group I right leg hams ($n=23$) were frozen at -18°C for 48 hours and thawed at 4°C . In Group II left leg hams ($n=23$) were refrigerated.

Both groups were cured separately with a mixture of salt (40 g/Kg of green ham) and nitrate in a ratio of 100:1. Fifteen days later the hams were washed and hung at $3-5^{\circ}\text{C}$ for 30 days, and then the temperature increases $1,5^{\circ}\text{C}$ weekly until the six month. A twelve cm ham slice from the middle of each ham was taken in order to evaluate white film.

100 g of ham from the medium part of Biceps femoris muscle was used for chemical analysis.

23 hams were sliced, three slices per ham has been stored at three different temperatures $3-4^{\circ}\text{C}$, $12-14^{\circ}\text{C}$, $22-24^{\circ}\text{C}$. White film has been evaluated after 15 days. Kramer test was used in order to compare the difference between tratements.

Sodium chloride was analyzed after the method of Charpentier-Volhard.

Protein content was analyzed after the Kjeldhal method. Tyrosine was determined after the technique of Wood, Sigurdsson and Dyer modified by Pearson (1968).

The Wilcoxon-Mann-Whitney Rank sums

test was used in order to compare non parametric data between high pH hams and low pH hams. The parametric data were analyzed by the analysis of variance.

RESULTS AND DISCUSSION

Group I hams has had a higher salt intake ($P < 0,05$) due to an acceleration in salt penetration in freeze-d hams, produced by the cellular disruption carried out by ice crystals.

In Group I the number of hams with crystals of tyrosine ($n=19$) was significantly higher ($P < 0,01$) than in Group II ($n=1$), and as shown in table II viceversa occurs for white film ($P < 0,05$).

In a similar form as the crystallization phenomenon in a solution, a high number of crystals would be found in the ham if the tyrosine concentration rises over the maximum solubility and a rapid supersaturation is produced by a rapid drying process (increasing the salt content and decreasing humidity) or a decrease in temperature. Freezing could disrupt cellular membranes and facilitate tyrosine diffusion to the surface of the crystal. As supersaturation decreases the rate of crystal formation gets down. Thus in hams as iberian serrano hams with an aging period of 18 months, the rate of crystal formation will be smaller, and the recrystallization during the broad period of aging will facilitate the growth of the large crystals.

Tyrosine precipitates have been found in the inner part of the bone only in group I hams.

The hams with a $pH > 6,2$ in the Semi-membranosus and Biceps femoris showed a small number of tyrosine crystals in freeze-d hams, a lower white film in refrigerated hams ($P < 0,01$) and a decrease in tyrosine concentration. It could be explicated by the increase in solubility as pH moves away from the isoelectric point ($pI = 5,63$). (Butz et. al., 1974; Kemp et.al., 1988) found no difference in white film samples stored at different temperatures.

In our experiment significant differences ($P < 0,01$) in white film have been found between different storage temperatures. The difference was more important when film scoring was < 3 . The lower the temperature the intenser the white film, owing to the decrease of the solubility of tyrosine as temperature goes down.

Table I

	Group I	SD	Group II	SD	
pHBF1	6,05	0,45	5,94	0,40	NS
pHBF2	6,43	0,21	6,30	0,23	NS
Humidity	61,19	3,23	62,20	2,65	NS
NaCl	7,21	0,97	6,30	1,12	*
Tyrosine(3)	284	113	305	106	NS
Proteine	28,15	4,17	27,00	2,57	NS
Hams with tyrosine crystals	19	-	1	-	

pHBF1 = pH in Biceps femoris in fresh ham

pHBF2 = pH in Biceps femoris at the end of the aging process

(3) : Tyrosine in mg/100 g

Table 2. Distribution of the white film scores

Group	film scores					
	0	1	2	3	4	5
Group I	0	8	9	6	0	0
Group II	0	3	7	5	5	3

Film scoring 0=none 1=very slight
2=slight 3=moderate 4=heavy
5=very heavy

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