The degradation of polyphosphate in meat matrix

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SUMMARY

Research has been conducted simulting conditions present in the meat transformation industry to affirm the degra dation of tripolyphosphate in the meat matrix. An investigation was also carried out under conditions similar t those found in the industry using a mixture of mortadella and wurstel as a substratum.

For the analysis it was adopted the ion exchange chromatography and the results demonstrated the following:

1 - The cause of the degradation is clearly enzymatic.

2 - All of the tripolyphosphate is degraded to orthophosphate within a shorter period of time than that wich <sup>is</sup> required for the product to reach the temperature that inactives the tripolyphosphatase.

3 - Pyrophosphate is the intermediate product of degradation.

4 - Any nutritional doubts about the presence of tripolyphosphates or pyrophosphates in cooked meat products are unfounded since only orthophosphates can be found.

## INTRODUCTION

Polyphosphates are additives widely used in several food industries and particularly in the meat industry in all countries.

Italy authorized the use of these additives in meat products with the D.M. of March 31, 1965 and it's <sup>succes;</sup> sive modifications in the following doses:

1) canned meat pies 0,25%

2) cooked shoulders and hams 0,25%

3) cooked salami (mortadella, wurstel, ect.) 0,40%

The chatacteristics of polyphosphates are, by this time, fully known (GHINELLI, 1976; ALLAVENA, et al. 1976). CANTONI, et al. 1976). However, there are differing opinions about how their mechanism restores the "power of water retention" of the muscle.

At first, the "hydration effect" of the polyphosphate was correlated with the increase of the pH-values (WIEKER, 1956), and then it was attributed to the complexing capabilities of certain cations. Presently, the propensity is to think that the true cause is the introduction into the meat matrix of the phosphoric ions con tained in these additives which perform the same function as ATP in the living cell (RUF, 1956).

In addition, it is known that polyphosphates, those with a high condensation degree (i.e. hexapolyphosp<sup>er</sup> tes) or those with a low condensation degree (i.e. pyrophosphates), are subject to hydrolysis eitherch<sup>emically</sup> with the help of acids, or enzymatically though specific enzymes naturally present in various food matrixes (polyphosphatases) or even characteristic of some bacteria (HAMM et al., 1977).

For this reason, a series of studies were conducted to verify the type of hydrolysis and it's veloc<sup>ity</sup> utilizing the most widely used polyphospate (sodium tripolyphosphate) in various meat matrixes and specifically in beef and pork meat, as well as in a pork mixture like that of mortadella and wurstel. MARERIALS AND METHODS

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The following meats were used in the experiments;

1) lean pork shoulder

2) lean beef shoulder

3) choice mortadella mixture (classic italian)

4) choice wurstel mixture (classic italian)

10 ml of tripolyphosphate acqueous solution (3.637%), corresponding the 2,50% of tripolyphosphate <sup>i0<sup>n5</sup></sup>

<sup>%s</sup> <sup>added</sup> to 50 g of finally gound meat matrix. Distilled water was then added until the total weight achieved  $^{96}$  100 g. In this way, the final concentration of product was equal to 0,50%.

At pre-arranged intervals, a 1 g sample was drawn, briefly homogenized with 100 ml of distilled water, and <sup>thered</sup> through a 0,45 micron membrane to obtain a 1-2 ml fluid portion.

The filtered sample was then analyzed with HPLC on an ionic exchange column equipped with conductometric <sup>Nevelation</sup>.

The experiments were divided into two series: the first was conducted by maintaining the samples at room <sup>Amperature</sup> (22-23°C) and the second by immersing the samples in a water-thermostat (75°C). The temperature at <sup>the CO</sup>re of the samples was recorded at the varieus drawing times with thermoeletric sensors. IPPARATUS

Nuppressor	Dionex mod 2.00 I
	Dionex mod AMMS I. P:N 038019
COLUMP	Dionex Conductivity
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Water solution NaOH 0.07 n <sup>Aegenerating</sup> phase 11 н NaHCO3 6.62 mN 11 п H2SO4 70.0 mN

## DERATING CONDITIONS

Wile phase efflent velocity And the settlent verocity 5-6 ml per min. 1.2 ml per min. AND DISCUSSION

The results are reported in the graphics from 1 to 4.

The experiments conducted on the beer and pork revealed a representation of the pork. Simultaneously, py-time they were added to the meat: 100 minutes for the beef and 120 minutes for the pork. Simultaneously, py-The experiments conducted on the beef and pork revealed a rapid disappearance of tripolyphosphates from <sup>the</sup> they were added to the meat: 100 minutes for the best from the time they were added to both meats. At this <sup>Point</sup> the pyrophosphates begin a gradual disappearance that is completed in approximately 6 hours.

In the thermostatically conducted experiments it is evidences and  $r_{e_s}$  reachs 50°C the hydrolysis of the tripolyphosphate is arrested. The residual values compared to the initial  $r_{e_s}$  to the hydrolysis of the tripolyphosphate is arrested. In the thermostatically conducted experiments it is evidenced that when the temperature of the meat sam-Alues are 28% for the beef meat and 36% for the pork.

The results of the experiments performed on the moreaders and developments. For example the disap-only difference being the length of time required to complete the same developments. For example the disap-The results of the experiments performed on the mortadella and wurstel proved to be quite similar, with <sup>harance</sup> of tripolyphosphates required 7 hours in the mortadella and 10 hours in the wurstel. The behavior of the <sup>hyronk</sup> <sup>by</sup> of tripolyphosphates required 7 hours in the mortadella and 10 hours in the the second s in the Wurstel. The disappearance of the pyrophosphates required 26 hours and 28 hours respectively.

The behavior of the same thermostatically samples confirmes where and a support of the same thermostatically samples confirmes where and a support of the meats. Namely that the hydrolysis of the two examined polyphosphates stopped when the temperature the The behavior of the same thermostatically samples confirmes what was pointed out in the previous experi-<sup>of the meats.</sup> Namery .... <sup>the mat</sup>rixes reached approximately 55 °C. CONCLUSIONS

- the tripolyphosphates begin their enzymatic hydrolysis immediately after the addition to the meat matrix and stop when the temperature reachs approximately 60°C in the meat samples and 55°C in the meat mixtures;

the hydrolysis velocity is the same for the two meats examined;

- simultaneous with the disappearance of the polyphosphate there occurs the appearance of the pyrophosph<sup>ate</sup> which reachs it's maximum content value at the time of the complete disappearance of the tripolyphosphate. Success sively, also the pyrophosphate hydrolyzes until it's complete disappearance requiring the same time for the two species examined;

- the relatively short times necessary for the hydrolysis of the tripolyphosphates warrant the hypothesis <sup>that</sup> the effects presently attributed to the polyphosphoric ion should be accredited merely to the phosphoric ions originated from the hydrolysis of the tripolyphosphate and pyrophosphate used;

the required times of hydrolysis, both in simple and complex meat matrixes, are fully within the technological times necessary for meat products (mortadella and wurtsel) before the temperature reachs the inactivation value of the enzyme. Consequently, the final product will not contain any significant quantity of the polyphosphate in any case, only 10-15% of the employed quantity;

- any doubt about the danger of ingetion of the polyphosphate is eliminated because the hydrolysis of the poly phopshate transforms it into orthophosphate, currently present in many foods and beverages and generally in hi gher quantities.

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