PACE OF BLOOD PLASMA ACIDIFICATION ON HAEME PIGMENT NITROSATION AND CURED COLOUR DEVELOPMENT

ASMA GEL /LIVEX/

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Vdrochloric acids on the curing efficiency of hemoglobine content in it was evaluated. It was found that acidification resulted in approx. 20% increase of haeme pigment conversion to nitrosopigment in comparison throl sample and that when the cured plasma was used for the livex processing the colour of the obtained gel was fairly stable and sensorically typical and characteristic of cured, cooked processed meat products.

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investigations on the hemoglobin curing and use of the resulting nitroso form for processed meat products purposes were carried out by several authors./MÖHLER and BAUMAN,1971, PALMIN at al 1973, SHAHIDI at al SHAHIDI and PEEG, 1988/.

authors' experience indicates that substantial increase of the hemoglobin reactivity with the nitrate i.e. pigment conversion to the nitrosopigments is possible since the changes in globin conformation resulted thermal denaturation, treatment by acids and/or enzymes. /JARMOLUK at al,1991/.

## TALS and METHODS

studies were performed on three separate batches of commercially manufactured pig blood plasma stabilized sodium citrate, frozen and defrosted at 4-6°C for 24 h, before further use. Thawed blood plasma was acidito pH=6.3 by 10% solutions of: lactic /A/, propionic /B/, acetic /C/, formic /D/ and hydrochloric /E/ acids. fied blood plasma was cured at 4-6°C for 1 hour using nitrite and sodium ascorbate in the molar ratio of sodium ascorbate: NaNO and hemine as 9:3:1. Samples of cured blood plasma were then destabilized according atented procedure /PATENT/ and using pig brain tissue homogenate as an enzymic activator /stimulator/ for pre-gelation. /DUDA at al,1989/. The resulted gels of plasma i.e. the raw livex formed was then pasteurized water bath at 80°C until 80°C was reached in the core of the sample. /DUDA and JARMOLUK,1985/. In each of the setitions of the study the following 7 variants of livex experimental batches were processed:

 $\frac{\log n}{1}$  not acidified and not cured plasma

2 - not acidified, cured plasma

- lactic acid acidified plasma, cured

- propionic acid acidified plasma, cured

C - acetic acid acidified plasma, cured

- formic acid acidified plasma, cured

 $oldsymbol{eta}_{oldsymbol{\xi}}$  - hydrochloric acid acidified plasma, cured

the obtained thermally denaturated plasma gel /livex/ the degree of haeme pigment conversion to the nitrosothe obtained thermally denaturated plasma gel /livex/ the degree of haeme pigment conversion to the nitrosothe obtained thermally denaturated plasma gel /livex/ the degree of haeme pigment conversion to the nitrosothe obtained thermally denaturated plasma gel /livex/ the degree of haeme pigment conversion to the nitrosothe obtained thermally denaturated plasma gel /livex/ the degree of haeme pigment conversion to the nitrosothe obtained thermally denaturated plasma gel /livex/ the degree of haeme pigment conversion to the nitrosothe obtained thermally denaturated plasma gel /livex/ the degree of haeme pigment conversion to the nitrosothe obtained thermally denaturated plasma gel /livex/ the degree of haeme pigment conversion to the nitrosothe obtained thermally denaturated plasma gel /livex/ the degree of haeme pigment conversion to the nitrosothe obtained thermally denaturated plasma gel /livex/ the degree of haeme pigment conversion to the nitrosothe obtained thermally denaturated plasma gel /livex/ the degree of haeme pigment conversion to the nitrosothe obtained with the following coefficients and by a standard with the following coefficients an

## RESULTS and DISCUSSION

An average value of defrosted plasma pH was 8.1 and was by approx. 0.5 pH units higher than the initial value of defrosted plasma pH was 8.1 and was by approx. The pH of the control livex samples /LK and LK2/ i.e. processed from not acidified plasma, after thermal tree fac ment, decreased to the level of pH=7.71, while processed from acidified plasma increased from the initial  $6^{,3}$ approx. 0.1 pH unit. Tab.1.

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Table 1. Influence of pig blood plasma acidification to pH=6.3 on gelation time of raw livexes, their pH after thermal treatment and on the degree of haeme pigment conversion to nitrosopigments /n=3/.

	Variant of livexes							
Parameters	LK <sub>1</sub>	LK <sup>2</sup>	LA	LB	LC	LD	LE	LSI
Н	7.71 <sup>b</sup>	7.71 <sup>b</sup>	6.40 <sup>a</sup>	6.42 <sup>a</sup>	6.39 <sup>a</sup>	6.40 <sup>a</sup>	6.41 <sup>a</sup>	0
Gelation time /s/	78.0 <sup>a</sup>	82.0 <sup>a</sup>	247.0°	251.0°	224.0 <sup>b</sup>	264.0 <sup>d</sup>	314.0 <sup>e</sup>	6
Degree of haeme Digment conversion /%	/ -	12.2 <sup>a</sup>	14.4°	13.9 <sup>b</sup>	15.1 <sup>d</sup>	15.2 <sup>d</sup>	14.9 <sup>d</sup>	0

Means with different superscripts are significantly different at p ≤0.05

The acidification of the plasma used for the experimental livex manufacturing, irrespectively of the  $ac^{\dot{1}\dot{0}}$ and in spite of using the enzymic activator of plasma gelling, caused about a triple delay in the dynamic fibrin net forming ofter plasma destabilization. Tab.1.

The lowering of the plasma pH below 6.0 resulted in practically total inhibition of the processes responsible for the fibrinogen transformation into fibrin and therefore processing of the livex from plasma acidified below 6.0 pH was impossible, indicating the importance of pH for fibrinogen transformation into fibrin.

An average content of the hemoglobin in plasma, expressed as hematin, amounted to 145 ppm. It was  $obser^{ved}$ that the degree of haeme pigment nitrosation to nitrosopigments was influenced by the accidification of plasopiawell as was dependent on the acid used. In the control sample i.e. not acidified /LK2/ the degree of haeme pill conversion to nitrosopigments was on average 12.2% while in livex processed from acidified plasma ranged  $f^{rol}$ 13.9% /LB/ to 15.2% /LD/. Tab.1.

In spite that the haeme pigment conversion was in acidified samples relatively small in comparison to  $t^{h^{\varrho}}$ control sample, the observed colour of livexes processed from acidified plasma was noticeably better. The determined values for  $L^*$  ,  $a^*$  and  $b^*$  livexes processed from acidified plasma i.e.: LA, LB, LC, LD and  $L^E$  were 68.5 ; 10.6 and 9.2 respectively and in comparison to the data determined for the control samples /LK<sub>2</sub>/ the observed colour parameters increased by 1.2; 2.8 and 2.0 respectively. Tab.2.

An average value for the hue determined for livexes manifactured from the acidified plasma and expressed in measure of angle were 10° smaller in comparison to control sample LK. Over 2 fold increase of the colour chro for livexes processed from acidified plasma, in comparison to control samples, was observed. Tab.2. Moreover colour of livexes processed from acidified plasma exhibited greater luminance and better chroma, when assets the colour of livexes processed from acidified plasma exhibited greater luminance and better chroma, when assets the colour of livexes processed from acidified plasma exhibited greater luminance and better chroma, when assets the colour of livexes processed from acidified plasma exhibited greater luminance and better chroma, when assets the colour of livexes processed from acidified plasma exhibited greater luminance and better chroma, when assets the colour of livexes processed from acidified plasma exhibited greater luminance and better chroma, when assets the colour of livexes processed from acidified plasma exhibited greater luminance and better chroma, when assets the colour of livexes processed from acidified plasma exhibited greater luminance and better chroma, when assets the colour of livexes processed from acidified plasma exhibited greater luminance and better chroma, when assets the colour of livexes processed from the colour of livexes processed fr visually and in comparison to the control sample.

The data obtained for physical parameters of colour for the illuminated samples i.e. for the colour stability are presented in Table 3. During sample illumination slight increase of colour luminance /L $^*$ / was observed. Simultaneously, substantial decrease of the red fraction of colour was notified and increase of the yellow in colour tone of illuminate. in colour tone of illuminated samples was observed. This was reflected in changes of the hue data from the level of about 38° for "O" time of illuminated samples was observed. of about 38° for "O" time of illumination to approx. 53° after 7 h of the experimental material illumination It was also determined that the illumination of livex samples substantially decreases its colour chroma,

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restactorily, the degree of conversion of the haeme pigment content in it to the nitrosopigments i.e. by approx.

Comparison to the data determined for the control samples. This facilitates livex manufacturing with sen-

desired pinkish colour typical of cured processed meat products.

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Variations in physical colour parameters of experimental material /livex/ as influenced by acids used for plasma acidification /n = 3/.

lour	THE STATE WHICH SHAPE WHICH SHAPE SHAPE SHAPE	L*	a*	b <sup>34</sup>	hue	chroma
	LK <sub>1</sub>	60.14 <sup>b</sup>	2.73 <sup>a</sup>	5.49 <sup>b</sup>	63.62 <sup>e</sup>	6.13 <sup>b</sup>
	LK <sub>2</sub>	59.39 <sup>a</sup>	3.77 <sup>b</sup>	4.55 <sup>a</sup>	51.93 <sup>d</sup>	6.05 <sup>a</sup>
int.	LA	68.97 <sup>e</sup>	10.25 <sup>c</sup>	8.94 <sup>c</sup>	41.56 <sup>b</sup>	13.67 <sup>c</sup>
ve <sub>X</sub>	LB	69.30 <sup>e</sup>	10.18 <sup>c</sup>	8.95 <sup>c</sup>	41.75 <sup>bc</sup>	13.63°
	LC	68.05 <sup>d</sup>	10.91 <sup>e</sup>	9.34 <sup>d</sup>	41.02 <sup>a</sup>	14.43 <sup>e</sup>
	LD	68.37 <sup>d</sup>	10.58 <sup>d</sup>	9.39 <sup>de</sup>	42.02 <sup>c</sup>	14.20 <sup>d</sup>
100	LE	67.68 <sup>c</sup>	10.99 <sup>e</sup>	9.41 <sup>e</sup>	41.04 <sup>a</sup>	14.53 <sup>f</sup>
		0.50	0.10	0.05	0.49	0.10

Weahs with different superscripts are significantly different at p  $\ll$ 0.05

Table 3.  $\label{eq:Variations} \mbox{Variations in physical colour parameters of illuminated experimental material-livex $/n = 3/. $}$ 

Colour parameters		L*	a*	b*	hue	chroma
	0	65.47 <sup>a</sup>	10.75 <sup>e</sup>	7.71 <sup>a</sup>	37.97 <sup>a</sup>	13.33 <sup>e</sup>
	1	65.77 <sup>ab</sup>	9.60 <sup>d</sup>	7.85 <sup>b</sup>	41.62 <sup>b</sup>	12.49 <sup>d</sup>
	3	66.04 <sup>bc</sup>	8.11 <sup>c</sup>	8.03 <sup>c</sup>	47.01 <sup>c</sup>	11.48 <sup>c</sup>
Illumination time /h/	5	66.24 <sup>cd</sup>	7.21 <sup>b</sup>	8.16 <sup>d</sup>	50.92 <sup>d</sup>	10.95 <sup>b</sup>
	7	66.41 <sup>d</sup>	6.77 <sup>a</sup>	8.29 <sup>e</sup>	53.13 <sup>e</sup>	10.76
LSD		0.30	0.90	0.05	0.37	0.74

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