

## Energy Metabolism in Post Mortem Muscle

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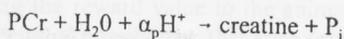
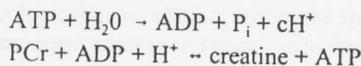
### INTRODUCTION

A dominant feature of biochemical events in post mortem muscle is the fall in pH. The rate and extent of the post mortem pH fall has important effects on meat quality, particularly when considered relative to the imposed chilling rate; the PSE condition in pork or cold shortening in beef are obvious consequences of, respectively, extremely fast and slow pH fall relative to temperature decline. The use of electrical stimulation has become an important processing tool for accelerating pH fall independent of temperature changes. There are therefore important commercial justifications for understanding and manipulating the mechanisms controlling energy metabolism in the post mortem muscle.

#### ATP changes in post mortem muscle

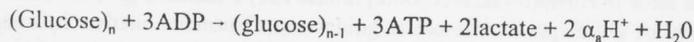
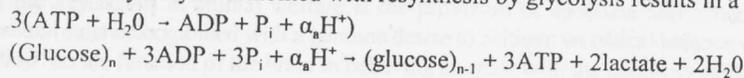
In the resting muscle, Bendall (1973) has proposed that the rate of ATP hydrolysis is primarily determined by myosin ATPase activity. On these grounds that the temperature dependence of post mortem pH decline is similar to the temperature dependence of myosin ATPase. In contrast, in the resting muscle, electrical stimulation increases intracellular calcium ( $[Ca^{2+}]_i$ ) and activates actomyosin ATPase to accelerate ATP turnover more than 100 fold. However, as much as 40% of the ATP use under these conditions can be attributed to non-contractile processes, primarily  $Na^+K^+$ -ATPase and  $Ca^{2+}$ -ATPase (Baker *et al.*, 1994).

Two separate mechanisms exist to regenerate ATP under the anaerobic condition of post mortem muscle, with differing effects on pH. The creatine kinase reaction catalyses the resynthesis of ATP from ADP and creatine phosphate (CP), one consequence of which is proton consumption:



where  $\alpha_a$  defines the proportion of produced  $\text{H}^+$  that is unbound, as defined by muscle buffering, and where  $\alpha_p = 1 - \alpha_a$ .

In contrast, ATP hydrolysis combined with resynthesis by glycolysis results in a net proton accumulation:



The majority of ATP resynthesis is therefore shared between two separate pathways. The pH decline in muscle is the sum of their respective contributions to  $[\text{H}^+]$ , together with changes in muscle buffering produced by the chemical state of the phosphate groups.

#### Electrical stimulation

Electrical stimulation is used to accelerate the rate of pH decline in post mortem muscle. Its effects on pH have two distinct components (Carr & Devine, 1978). First, there is the decline produced directly by the muscular contractions, from the accelerated energy expenditure associated with the high levels of work done by the muscle. Second, subsequent to the electrical stimulation, the rate of pH fall is increased, so that, after stimulation, the rate following stimulation is 50-70% greater than unstimulated control muscle. There remains therefore a sustained increase in the rate of glycolysis, and therefore of ATPase activity, after the muscle returns to the resting state following stimulation, but the causes of this increase remain to be fully described.

#### $^{32}\text{P}$ -NMR measurements of post mortem energy metabolism

$^{32}\text{P}$ -NMR is an effective technique for monitoring the energy metabolism in muscle (Meyer *et al.*, 1982; Vogel *et al.*, 1985; Renou *et al.*, 1985). It allows a continuous and non-invasive measurement of changes in the concentrations of high energy phosphate compounds (ATP, CP, inorganic phosphate ( $\text{P}_i$ ) and hexose phosphate), and this information can be used to assess post mortem energy metabolism. In the experiments described here, strips of lamb m. semitendinosus were prepared with minimal damage to the fibres, and recordings made from the distal portion of the muscle. The muscles were mounted onto stainless steel wire guides and inserted into 10 mm diameter recording tubes. Anaerobic conditions were maintained by immersion in mineral oil.  $^{32}\text{P}$ -NMR recordings were made using a Bruker AC300, with a multinuclear probe tuned for  $^{32}\text{P}$  at 125 MHz, and a  $60^\circ$  tip angle with 1 second relaxation delay. The peaks were quantified to  $\mu\text{M/g}$  wet weight according to the method of Vogel *et al.* (1985). pH changes in the muscle were calculated from the shift of the inorganic phosphate peak. Unless stated otherwise, all recordings were carried out at a constant temperature of  $35^\circ\text{C}$ .

The muscle samples were stimulated within the recording tubes using the wire guides as electrodes. A constant current stimulator was used, set to 50 mA, and a waveform of 10 ms pulses of alternating polarity at 15 Hz. The duration of stimulation was 60 seconds.

### Post mortem energy metabolism in muscle held at 35°C.

Figure 1 shows the changes in the concentrations of phosphate compounds from a representative muscle. It follows the general pattern described in other species using biochemical determinations (Bendall, 1979) and  $^{32}\text{P}$ -NMR (Vogel *et al.*, 1985; Renou *et al.*, 1986; Azuma *et al.*, 1994). A consistent difference was the absence of a delay phase (when ATP concentrations remain essentially constant; Bendall, 1973), as reported in previous studies. However, the delay phase was evident when the same muscle is held at 15°C, suggesting that the faster rate of ATPase activity at 35°C is responsible for the early onset of the decay phase.

Lactate accumulation can be used to estimate the rate of glycolysis, but cannot be measured directly from  $^{32}\text{P}$ -NMR. However, a close and linear relationship has been recognised between lactate concentration in muscle and pH (Bendall, 1973; Fabiansson & Reuterswärd), a relationship that in lamb is unaffected by electrical stimulation or by temperature (unpublished observation), and offers therefore a simple conversion using the measured value of 58  $\mu\text{M}$  lactate/pH/g wet weight. An alternative approach is to calculate the rate of glycolysis from changes in  $[\text{H}^+]$ , if the buffering of muscle is accounted for. By titration of homogenised post rigor muscle to measure buffering, and calculating the effects of phosphate metabolism (hydrolysis of CP and ATP) on the total buffering, the calculated rate of glycolysis was equivalent to the much simpler pH-lactate relationship.

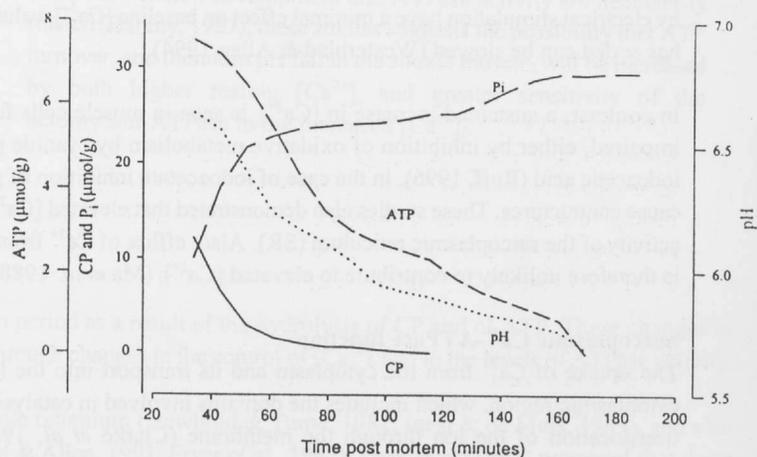


Figure 1.  $^{32}\text{P}$ -NMR results from post mortem muscle.

The rate of ATPase activity during the pre rigor period can then be described from the NMR data. The rate of resynthesis of ATP ( $\mu\text{M/g}$  wet weight/minute) can be calculated from the stoichiometry:

$$\Delta\text{ATP} = \Delta\text{PC} + \Delta(1.5 \text{ lactate}) + \Delta(2\text{ATP})$$

Figure 2 shows the rate of ATP turnover from a representative muscle during the pre-rigor period, expressed relative to pH rather than time post mortem. The rate of ATP turnover decreases progressively as the pH falls to pH 6.3, becomes stable until pH around 6.1 and then declines thereafter until rigor is complete. This result contrasts somewhat with those described by Bendall (1973), who noted a resurgence in ATP turnover in the intermediate pH range. The difference is probably associated with the earlier fall in ATP levels in lamb muscle at 35°C (the absence of a delay phase), since the resurgent pattern was evident in muscle analysed at 15°C (data not shown). Because ATP resynthesis can occur through either the creatine kinase reaction or by glycolysis, the proportional contribution of glycolysis to the total resynthesis during the pre rigor period is also shown in Figure 2. In the early post mortem period, when CP levels are still high, glycolysis contributes around 65% of the ATP resynthesis, but gradually rises to reach more than 90% once creatine phosphate is depleted. The remainder of the ATP activity is derived from the myokinase reaction.

Figure 3 shows the calculated rate of ATP turnover from a representative muscle following electrical stimulation. Overall, the rate of ATP turnover is increased by around 55% compared with unstimulated. As in unstimulated muscle, the rate of ATP turnover declines as the pH falls, but the decline is now much more pronounced. In contrast to unstimulated muscle, the rate of ATP turnover increases from pH 5.9 until rigor.

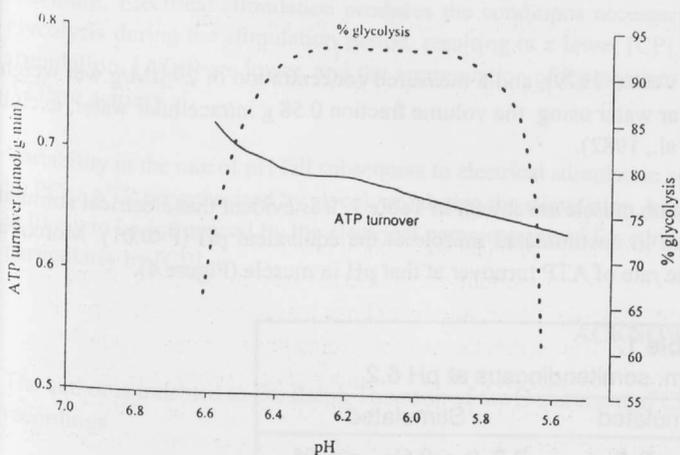


Figure 2. ATP turnover and % glycolysis in unstimulated muscle

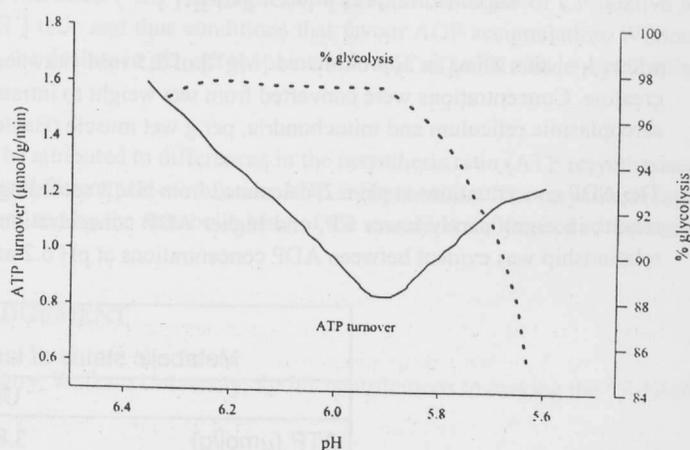


Figure 3. ATP turnover and % glycolysis in stimulated muscle

At an equivalent pH, CP levels are lower in stimulated muscle than in unstimulated muscle (Table 1). This means that the resynthesis ratio of resynthesised ATP derived from the creatine kinase reaction versus glycolysis, is increased during electrical stimulation. Results showing high levels of PCr contribution to total ATP resynthesis in the early phases of an electrical stimulation protocol have been reported in both anaerobic and aerobic muscle (Spriet *et al.*, 1987; Krause & Wegener, 1996)

#### Why is the rate of glycolysis accelerated in muscle following electrical stimulation?

While the pH fall during stimulation can be anticipated from the intense muscular activity, the cause of the accelerated rate of pH fall subsequent to the stimulation has not been effectively explained. The increased ATPase activity probably implicates elevated  $[Ca^{2+}]_i$ . Using  $Ca^{2+}$ -sensitive fluorescent probe introduced into oxygenated fibres, the measured resting levels of  $[Ca^{2+}]_i$  increase slightly when the muscle pH falls (Westerblad & Allen, 1993; Ashley, 1978; Westerblad & Allen, 1993), for example from 26 to 40 nM in single mouse muscle as the pH falls from 7.3 to 6.8 (Westerblad & Allen, 1993), but such changes in concentration are probably too small to generate significant metabolic effects. More substantial pH falls induced by electrical stimulation have a minimal effect on baseline  $[Ca^{2+}]_i$  values, although the rate of recovery of  $[Ca^{2+}]_i$  to baseline values once stimulation has ended can be slowed (Westerblad & Allen 1994).

In contrast, a sustained increase in  $[Ca^{2+}]_i$  is seen in muscle cells following electrical stimulation under conditions where ATP resynthesis is impaired, either by inhibition of oxidative metabolism by cyanide poisoning (Westerblad & Allen, 1991), or by inhibition of glycolysis by iodoacetic acid (Ruff, 1996). In the case of iodoacetate inhibition of glycolysis, electrical stimulation produces a sufficient increase in  $[Ca^{2+}]_i$  to cause contractures. These studies also demonstrated that elevated  $[Ca^{2+}]_i$  does not depend on depletion of ATP, that might impair the  $Ca^{2+}$ -ATPase activity of the sarcoplasmic reticulum (SR). Also, efflux of  $Ca^{2+}$  from the ryanodine channel is virtually absent when the pH falls below 6.2, it is therefore unlikely to contribute to elevated  $[Ca^{2+}]_i$  (Ma *et al.* 1988).

#### Sarcoplasmic $Ca^{2+}$ -ATPase function

The uptake of  $Ca^{2+}$  from the cytoplasm and its transport into the lumen of the SR involves an ATPase protein with two distinct regions: a cytoplasmic region, which includes the domains involved in catalysis, and a transport domain containing the  $Ca^{2+}$ -binding sites involved in translocation of the ion through the membrane (Clarke *et al.*, 1989). During catalysis, the ATPase protein cycles between two conformational states: a high  $Ca^{2+}$  affinity-low saturation (E1) state that binds to cytoplasmic  $Ca^{2+}$ ; and a low affinity-high saturation state that releases the bound  $Ca^{2+}$  into the lumen of the SR (E2) (de Meis & Vianna, 1979). The sequence of events that lead to transport of  $Ca^{2+}$  into the SR is however reversible, and can therefore mediate the efflux of  $Ca^{2+}$  from the SR into the sarcoplasm. In the presence of substrate (ADP and  $P_i$ ), the reaction can be coupled to the resynthesis of ATP (de Meis & Tume 1977; Tanford, 1984) but can also proceed without coupling to ATP synthesis (de Meis and Inesi, 1992). Under normal circumstances, the low affinity of the E2 site will ensure that the rate of the backwards reaction will be high in spite of high intraluminal  $[Ca^{2+}]$ , but it is evident that the net sarcoplasmic  $[Ca^{2+}]$  will be defined by the rate constants of the forward and backward reactions of the  $Ca^{2+}$ -ATPase.

#### Can the kinetics of SR $Ca^{2+}$ -ATPase account for increased $[Ca^{2+}]_i$ and accelerated glycolysis following electrical stimulation?

In order for there to be an increase in  $[Ca^{2+}]_i$ , the kinetics of the reverse reaction ( $Ca^{2+}$ -efflux) need to be enhanced relative to the forward reaction. Other than by an increase in SR  $[Ca^{2+}]$ , an unlikely consequence of electrical stimulation, experimental evidence has shown that efflux is stimulated by increased [ADP], particularly when SR  $[Ca^{2+}]$  are high (Pick & Bassilian, 1983; Inesi & de Meis, 1989). Under conditions where intraluminal  $[Ca^{2+}]$  between 20-40 mM, the rate of  $Ca^{2+}$  efflux is doubled by an increase in [ADP] from 1 to 100  $\mu$ M, an effect attributed to an increase in the proportion of nucleotide bound intermediates of the enzyme (Inesi & de Meis, 1989). Unfortunately, chemical measurements of [ADP] in muscle are difficult because the majority of ADP present in the cell is bound to myofibrillar proteins, particularly actin. It is therefore problematic, [ADP] in muscle are too low to be detected directly by  $^{32}P$ -NMR. However, free [ADP] can be calculated from the equilibrium constant of the CK reaction:

$$k = \frac{[Creatine][ATP]}{[ADP][CP][H^+]}$$

where  $k = 1.66 \times 10^9$  at an approximated  $[Mg^{++}]$  of 2.5 mM (Lawson & Veech, 1979), and a measured concentration of 29  $\mu$ M/g wet weight of creatine. Concentrations were converted from wet weight to intracellular water using the volume fraction 0.58 g intracellular water, excluding sarcoplasmic reticulum and mitochondria, per g wet muscle (Baylor *et al.*, 1982).

The ADP concentrations at pH 6.2, calculated from NMR recording in lamb muscle are shown in Table 1. It is evident that electrical stimulation results in significantly lower CP, and higher ADP concentrations than in unstimulated muscle at the equivalent pH ( $P < 0.01$ ). More importantly, a relationship was evident between ADP concentrations at pH 6.2 and the rate of ATP turnover at that pH in muscle (Figure 4).

	Unstimulated	Stimulated
ATP ( $\mu$ mol/g)	3.8 (s.e. 0.4)	2.9 (s.e. 0.6); $p > 0.05$
CP ( $\mu$ mol/g)	2.9 (s.e. 0.2)	0.9 (s.e. 0.07); $p < 0.01$
Calculated ADP ( $\mu$ M)	0.37 (s.e. 0.03)	0.76 (s.e. 0.04); $p < 0.01$

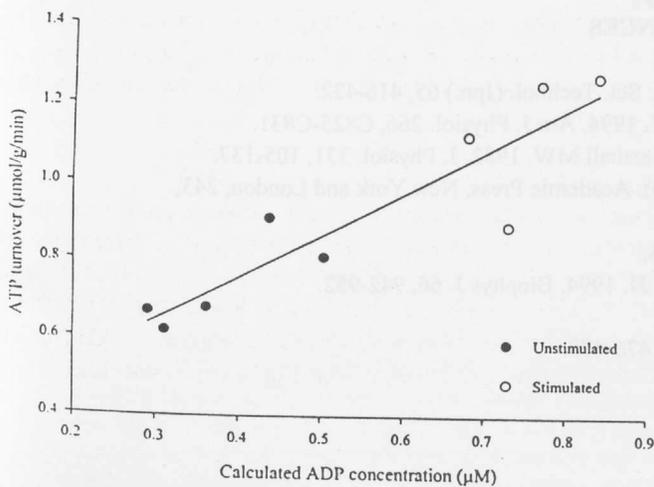


Figure 4. Relationship between calculated [ADP] and ATP turnover

### Myofibrillar sensitivity to $Ca^{2+}$

While changes in the kinetics of sarcoplasmic  $Ca^{2+}$ -ATPase may account for the observed rise in  $[Ca^{2+}]_i$  following stimulation of anoxic muscle, it still remains to be determined whether this is a sufficient explanation for the accelerated rate of pH fall in electrically stimulated muscle. Recently, Ruff (1996) has suggested that the myofibrillar sensitivity to  $Ca^{2+}$  is enhanced in the presence of elevated [ADP] (although other have not found such an effect, for example Cooke & Pate, 1985), resulting in a significant increase in the tension- $Ca^{2+}$  relationship of single mouse muscle fibres. On the basis that tension development and ATPase activity are necessarily related (Barany, 1967), these studies suggest the possibility that ATP turnover, and therefore pH fall in the anoxic muscle, will be increased by both higher resting  $[Ca^{2+}]_i$  and greater sensitivity of the actomyosin-ATPase to the increased  $[Ca^{2+}]_i$ .

### Role of $P_i$ in post mortem metabolism

Inorganic phosphate concentrations accumulate during the post mortem period as a result of the hydrolysis of CP and of ATP. These changes in  $[P_i]$  can have important influences on post mortem energy metabolism through changes in the control of  $[Ca^{2+}]_i$  and in the levels of ATPase activity.

Increasing  $[P_i]$  enhances the uptake capacity of  $Ca^{2+}$  into the sarcoplasmic reticulum (Newbold & Tume, 1981; Inesi & de Meis, 1989), and also reduces the release of  $Ca^{2+}$  from the sarcoplasmic reticulum (Westerblad & Allen, 1991; Fryer *et al.*, 1995). Fryer *et al.* (1995) proposed that these consequences of increased  $[P_i]$  could be accounted for by the accumulation of  $P_i$  within the sarcoplasmic reticulum, where it complexes with  $Ca^{2+}$  and thus reduces free  $[Ca^{2+}]_i$  in the sarcoplasmic reticulum; indeed, the solubility product of calcium phosphate can be exceeded within the range of  $[P_i]$  found in post mortem, resulting in precipitation of  $Ca^{2+}$ . Therefore, increased  $[P_i]$  reduces sarcoplasmic reticular  $[Ca^{2+}]_i$ , which stimulates uptake and inhibits efflux of  $Ca^{2+}$ , and so contribute to lower  $[Ca^{2+}]_i$ .

Increasing  $[P_i]$  also reduces actomyosin ATPase activity (Cooke & Pate, 1985; Parkhouse, 1991) and inhibits myofibrillar  $Ca^{2+}$  sensitivity (Fryer *et al.*, 1995). Similar effects on myofibrillar ATPase activity and  $Ca^{2+}$  sensitivity are produced by a drop in pH (Blanchard *et al.*, 1984; Parkhouse, 1991). From the standpoint of understanding the rate of post mortem ATP turnover and pH decline, the effect of increased  $[P_i]$  will be to oppose the influence of increasing [ADP] proposed earlier, by increasing uptake and reducing release of  $Ca^{2+}$  from the sarcolemma, and reducing the myofibrillar sensitivity to  $Ca^{2+}$ .

### CONCLUSIONS

Electrical stimulation of carcasses to accelerate pH fall offers important processing advantages, although, particularly in the case of low voltage stimulation, the accelerated pH fall subsequent to the stimulation can be variable (Eikelenboom *et al.*, 1985). Not unexpectedly,  $^{32}P$ -NMR data found that the rate of ATP turnover is increased following electrical stimulation and therefore drives the faster rate of pH decline. The link between the rate of ATP turnover and the calculated free ADP concentrations is proposed to be ADP-induced efflux of  $Ca^{2+}$  from the sarcoplasmic reticulum. Electrical stimulation produces the conditions necessary for increased [ADP] through the disproportionate use of CP relative to glycolysis during the stimulation period, resulting in a lower [CP] to  $[H^+]$  ratio and thus conditions that favour ADP accumulation. Without stimulation, [ADP] are lower, and the accumulation of  $P_i$ , together with the decline in pH, act to maintain low  $[Ca^{2+}]_i$  and reduce myofibrillar ATPase activity.

Variability in the rate of pH fall subsequent to electrical stimulation could be attributed to differences in the resynthesis ratio (ATP resynthesised by PCR / ATP resynthesised by glycolysis) during the stimulation, leading to different post-stimulation ADP concentrations. The resynthesis ratio is likely to be influenced by the electrical parameters used for stimulation, and by the metabolic state of the muscle at the time of stimulation, particularly by [CP].

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