### D1-21

#### APPLICATIONS OF COLD SET WHEY PROTEIN GELATION IN IMPROVING MEAT BATTERS

Hongsprabhas, P. and Barbut, S. Food Science Department University of Guelph Guelph, Ontario N1G 2W1 Canada

#### Abstract

The effect of using pre-heated whey proteins to replace part of the proteins in poultry meat batters formulated with different salt levels studied. Substitution with 2% pre-heated whey proteins significantly (P<0.05) improved binding in raw batters. In the cooked state, it entitle water holding capacity, reduced cook loss and improved textural parameters, especially at  $\leq 1.5\%$  salt. Un-heated whey proteins (i.e., lacking ability to get at low temperature) did not a be ability to get at low temperature) did not show any negative effect on the texture of the cooked batters, but reduced water holding capacity of raw batters. raw batters.

Overall, the study demonstrated the potential new use of whey proteins to improve the binding of raw and cooked meat batters, participut in reduced salt meat products.

#### Introduction

Salt-soluble proteins (myosin and actin) are the main components responsible for binding and texture in processed meat products. adequate salt level and proper mixing are essential to obtain optimal water retention and acceptable mouth feel. Non-meat ingredients (prot carbohydrates) which are often added to processed meat products can also affect the type of gel formed, and consequently the texture of the product. Overall, non-meat proteins are used in most formulations and affect the type of gel formed, and consequently the texture of the product. product. Overall, non-meat proteins are used in meat formulations to improve yield, modify textural properties and reduce costs (Hung and Support 1993). Proteins such as solve and white or when one with the texture of tex 1993). Proteins such as soy, egg white or whey are usually added to enhance the gel structure of meat products; however, sometimes can end negative effects. Lanier (1991) reported that substituting soy protein for turkey or surimi increased the cohesiveness of the gels, but slightly drop the rigidity at high replacement levels. Substitution with when we the construction of the gels of the gels of the gels of the gels of the gels. the rigidity at high replacement levels. Substitution with whey proteins (0.74%) increased gel cohesiveness of a surimi system. However, the whey levels (2.96%) resulted in higher rigidity and levels at the proteins (0.74%) increased gel cohesiveness of a surimi system. whey levels (2.96%) resulted in higher rigidity and lower strain. Burgarella *et al.* (1985) indicated that using egg white and WPC diluted interfered with the selation of the fish proteins in surgini. The interfered with the gelation of the fish proteins in surimi. Their mechanical testing suggested that the initial work values obtained for either the and egg white gels or fish and whey mixtures were lower than those for each single protein system, mainly because of a dilution effect of the protein by one another. Another suggestion was that when and each single protein system, mainly because of a dilution effect of the protein system. by one another. Another suggestion was that whey and egg white proteins became reactive at temperatures above the gelation temperature of the protein and, therefore, the mixed cells are weaker. fish protein and, therefore, the mixed gels are weaker. Overall, a better understanding of the contribution of non-meat proteins to the text forming process of meat protein cole is important. forming process of meat protein gels is important.

Previous work has indicated that a pre-heated WPI suspension could gel at a low protein concentration (2% w/v) and low temperature the presence of >20 mM  $Ca^{2+}$  (Honosprehen and Party 1007). On the protein concentration (2% w/v) and low temperature to the presence of >20 mM  $Ca^{2+}$  (Honosprehen and Party 1007).  $3^{\circ}$ C) in the presence of >20 mM Ca<sup>2+</sup> (Hongsprabhas and Barbut, 1997). Gels formed at low temperature and [CaCl<sub>2</sub>] are more transparent loss rigid than these formed at higher temperature and [CaCl<sub>2</sub>] are more transparent. less rigid than those formed at higher temperature and  $[CaCl_2]$  above 30 mM CaCl<sub>2</sub>. Thus, it is possible that Ca<sup>2+</sup>-induced gelation of  $WPI_{vl}$ be beneficial in meat systems as it can provide binding prior to cooking. Furthermore, the mechanisms of WPI gelation could be manipulated medifying the ionic strength of the system and here the modifying the ionic strength of the system and hence the gel characteristics. Thus, the objective of this study was to examine the effect of pre-WPI and NaCl concentration on the properties of poultry meat batters.

#### **Materials and Methods**

Hand-deboned chicken breast meat was trimmed of all visible fat and connective tissue and chopped in a bowl chopper to produce homogeneous mass. The meat was vacuum packed and frozen for up to one month prior to use.

Whey protein isolate (Davisco, MN) suspensions (10%) were prepared in double distilled water and half was pre-heated at  $80^{\circ}C_{1}^{\circ}$ min and then cooled at room temperature. The other half was kept without heating. The meat was thawed at 1°C for 24 hr, portioned and mit with the other ingredients with the other ingredients.

Salt (NaCl; 0-2%) was added and mixed by hand for 1 min followed by the addition of water or WPI suspensions (pre-heated or non-heated or nonand mixed for another 4 min. Treatments consisted of: (i) whey protein: meat protein = 0.16% (control); (ii) pre-heated whey protein: meat protein = 2.14% and (iii) non-heated whey protein: meat protein = 0.16% (control); (ii) pre-heated whey protein: meat protein = 0.16% (control); (iii) pre-heated whey protein: meat protein = 0.16% (control); (iii) pre-heated whey protein: meat protein = 0.16% (control); (iii) pre-heated whey protein: meat protein = 0.16% (control); (iii) pre-heated whey protein: meat protein = 0.16% (control); (iii) pre-heated whey protein: meat protein = 0.16% (control); (iii) pre-heated whey protein: meat protein = 0.16% (control); (iii) pre-heated whey protein: meat protein = 0.16% (control); (iii) pre-heated whey protein: meat protein = 0.16% (control); (iii) pre-heated whey protein: meat protein = 0.16% (control); (iii) pre-heated whey protein: meat protein = 0.16% (control); (iii) pre-heated whey protein: meat protein = 0.16% (control); (iii) pre-heated whey protein: meat protein = 0.16% (control); (iii) pre-heated whey protein: meat protein = 0.16% (control); (iii) pre-heated whey protein: meat protein = 0.16% (control); (iii) pre-heated whey protein: meat protein = 0.16% (control); (iii) pre-heated whey protein: meat protein = 0.16% (control); (iii) pre-heated whey protein: meat protein = 0.16% (control); (iii) pre-heated whey protein: meat protein = 0.16% (control); (iii) pre-heated whey protein: meat protein = 0.16% (control); (iii) pre-heated whey protein: meat protein = 0.16% (control); (iii) pre-heated whey protein: meat protein = 0.16% (control); (iii) pre-heated whey protein: meat protein = 0.16% (control); (iii) pre-heated whey protein: meat protein = 0.16% (control); (iii) pre-heated whey protein: meat protein = 0.16% (control); (iii) pre-heated whey protein = 0.16% = 2:14%; and (iii) non-heated whey protein: meat protein = 0:16% (control); (ii) pre-heated whey protein: meat protein = 2:14%. Portions of the mixtures were stuffed into 50 mL test tubes, which were divergent into two sets. The first was used for a pre-application exclusion of the mixtures were stuffed into 50 mL test tubes, which were divergent into two sets. into two sets. The first was used for a pre-cooking evaluation and the second for a post-cooking evaluation (78°C internal temperature). sets were kept overnight (16 hr) at 1°C to allow for cold-set gelation of the WPI.

Water holding capacity (WHC) was evaluated by centrifugation at 15,300g for 20 min. Supernatant was collected and WHC calculated of the second as the proportion of liquid retained. Colour was determined by a Minolta Chroma Meter CR-200b. Penetration force was determined by a plant diameter stainless stal rad memory of the plant diameter state diame diameter stainless steel rod mounted on a Nene Texture Analyzer descending into the meat samples. Cook loss was determined as the propo of liquid separating from the cooked batters.

#### **Results and Discussion**

The substitution of WPI, either in the pre-heated or non-heated form, significantly ( $P \le 0.05$ ) affected WHC and textural properties  $\binom{1}{k}$ 1). It also affected colour and cook loss of the raw and cooked meat batters. Salt concentration significantly affected WHC, colour, cook and penetration force of the raw and cooked batters.

Water holding and penetration force markedly increased when pre-heated WPI suspensions were used to substitute for part of the chile meat protein (Table 1). Although the non-heated WPI substitution consisted of the same amount of protein as the other two treatments structures of the non-heated WPI batters were significantly weaker. When no salt was added, no penetration force was detected because the ball was too soft. In addition, WHC of the same an and the same and the same and the same addition of the same and the same addition when the same addition of the same addition when the same addition of the same addition when the same addition when the same addition when the same addition of the same addition when the s was too soft. In addition, WHC of the samples were much lower than that of treatments (i) and (ii).

It should be noted that a solution of 2% pre-heated WPI does not form a gel even in the presence of 10 mM CaCl<sub>2</sub> when incubated at <sup>1/2</sup> for 4 days. Thus, the structure formed within the meat batter prepared with 0% NaCl was possibly due to an interaction between the WPI and more than the presence of a literation between the WPI and more than the presence of a literation between the WPI and more than the presence of a literation between the WPI and more than the presence of a literation between the WPI and more than the presence of a literation between the WPI and more than the presence of a literation between the WPI and more than the presence of a literation between the WPI and more than the presence of a literation between the WPI and more than the presence of a literation between the WPI and more than the presence of a literation between the WPI and more than the presence of a literation between the WPI and more than the presence of a literation between the WPI and more than the presence of a literation between the WPI and more than the presence of a literation between the WPI and more than the presence of a literation between the WPI and the presence of a literation between the WPI and the presence of a literation between the WPI and the presence of a literation between the WPI and the presence of a literation between the work of a proteins in the presence of salt naturally occurring within the muscle. It is possible that the pre-heated whey proteins, which have more read groups than the non-heated ones, could react with the soluble meat proteins (present in the aqueous phase) of the meat batters.

As NaCl was raised, WHC increased and the meat became significantly darker. Replacing 2% of the meat proteins with pre-heated w did not affect the lightness of the batter at salt level <2%, but replacement with non-heated WPI resulted in a significantly darker appearance

# all the salt levels but the 2%.

Within the all-meat batters and the ones substituted with non-heated WPI, the increase in salt slightly increased the penetration force at the Within the all-meat batters and the ones substituted with non-heated wPI, the increase in said slightly instead of forming a cold-set gel. How stage (Table 1). In the pre-heated WPI group, the meat batters became significantly more firm. This was the result of forming a cold-set gel. However, increasing the salt level resulted in decreasing the penetration force indicating that higher salt levels produce softer gel structures. This  $w_{as}$  expected since increasing the Na level in a WPI cold-set system is known to shift the gel structure from fine strands to aggregates (Hope in the structure from the structure from the structure heated WPI, reduced the Hongsprabhas and Barbut, 1996). Thus, the results indicate that increasing the salt in raw meat batters, containing pre-heated WPI, reduced the Penetration and Barbut, 1996). Thus, the results indicate that increasing the salt in raw meat batters, containing pre-heated WPI, reduced the penetration force. In most processed meats, salt is required to extract salt-soluble proteins such as myssin, actin, acting stratign force. their amounts in the aqueous phase are proportional to the NaCl used (Gordon and Barbut, 1992). In the present study, the lower penetration force of samples containing pre-heated WPI with higher salt levels might have resulted from changes in the charge distribution on the WPI molecules which caused more aggregation or the higher salt level resulted in more meat soluble proteins which have interfered with the WPI gel network formation.

## References

ngi

refo

otell e fil

Smil exhi

ropp high

ited

het rotel

of cxt

urc

nta WO ated

for m

1d m

anci

Burgarella, J.C., Lanier, T.C. and Hammann, D.D. 1985. Effects of added egg white or whey protein concentration on thermal transitions in rigidity of croaker surimi. J. Food Sci. 50:1588.

Gordon, A. and Barbut, S. 1992. Effect of chloride salts on protein extraction and interfacial protein film formation in meat batters. J. Sci. Food Agric. 58:227.

 $H_{0ngsprabhas}$ , P. and Barbut, S. 1996. Ca<sup>2+</sup> induced gelation of whey protein isolate: effect of pre-heating. Food. Res. Int. 29:135.

Hongsprabhas, P. and Barbut, S. 1996. Ca<sup>2+</sup> induced gelation of whey protein isolate. Check of pro-inducing protein isolate. Lebensm.-Wiss u.-Technol. 30:45.

Hung, T.-Y. and Smith, D.M. 1993. Dynamic rheological properties and microstructure of partially insolubilized whey protein concentrate and chicken breast salt-soluble protein gels. J. Agric. Food. Chem. 41:1372.

chicken breast salt-soluble protein gels. J. Agric. Food. Chem. 41:1372. R. Barford (Eds), p.268. ACS Series 454, Washington, D.C.

<sup>K</sup>. Barford (Eds), p.268. ACS Series 151, ... Cary, NC. Shill 990. User's Guide: Statistics. SAS Institute. Cary, NC.

Smith, D.M. 1991. Factors influencing heat-induced gelation of muscle proteins. In Interactions of Food Proteins. N. Parris and R. Barford (Eds), p.243. ACS Series 454, Washington, D.C.

Table 1.

Effect of WPI substitution and NaCl concentration on functional properties of raw and cooked chicken breast meat batters. Total protein content in all batters was 16%.

NaCl (%)	WHC (%)	Raw penetration force (N)	Cooked penetration force (N)
		See Manager Provide	2 oct
0.0	66.5 <sup>g</sup>		3.06 <sup>d</sup>
1.0	89.3 <sup>d</sup>		5.52 <sup>cd</sup>
1.5	99.1 <sup>ab</sup>	0.54 <sup>cd</sup>	6.10 <sup>cd</sup>
2.0	99.3 <sup>ab</sup>	0.73°	6.10 <sup>cd</sup>
0.0	83.7°	2.96ª	7.99 <sup>abc</sup>
		2.13 <sup>b</sup>	10.29 <sup>ab</sup>
			10.71ª
2.0	99.8ª	2.11 <sup>b</sup>	8.02 <sup>abc</sup>
	distant of	Beweite Boilty Hilling	2.014
0.0			2.91
1.0	78.6 <sup>r</sup>		6.32°
1.5	88.2 <sup>d</sup>		6.83°
2.0	92.5°	0.41 <sup>cde</sup>	7.25 <sup>bc</sup>
	0.0 1.0 1.5 2.0 0.0 1.0 1.5 2.0 0.0 1.0 1.5	$\begin{array}{cccccccccccccccccccccccccccccccccccc$	NaCl (%)WHC (%)penetration force (N) $0.0$ $66.5^{g}$ $0.27^{de}$ $1.0$ $89.3^{d}$ $0.35^{cde}$ $1.5$ $99.1^{ab}$ $0.54^{cd}$ $2.0$ $99.3^{ab}$ $0.73^{c}$ $0.0$ $83.7^{e}$ $2.96^{a}$ $1.0$ $97.4^{b}$ $2.13^{b}$ $1.5$ $99.3^{ab}$ $2.14^{b}$ $2.0$ $99.8^{a}$ $2.11^{b}$ $0.0$ $53.7^{h}$ $na^{*}$ $1.0$ $78.6^{c}$ $0.10^{e}$ $1.5$ $88.2^{d}$ $0.25^{de}$

 $M_{\text{eans}}$  followed by a different superscript are significantly different (p $\leq 0.05$ ). <sup>not</sup> available; batters too soft.